Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2023-24 Semester: Even

Name of Assistant Prof./Associate Prof./Professor: Dr. Ravinder Singh

Class: B.Sc. 2<sup>nd</sup> /4<sup>th</sup> /6<sup>th</sup> Semester (Chem. Hons.)

| 16 <sup>th</sup> January 2023 to 16 <sup>th</sup> | May 2023 [B.Sc. 2 <sup>nd</sup> Semester Chemistry Hons.] Physical Chemistry |
|---|--|
| Week 1  | Chemical Kinetics  |
| 16 <sup>th</sup> January – 20 <sup>th</sup>       | Chemical kinetics and its scope, Rate of reaction, factors influencing the   |
| January   | rate of reaction. Concentration, temperature, pressure, solvent, light,      |
|   | catalyst, concentration dependence of rates.                                 |
| 21st January                                      | SUNDAY   |
| ,   |  |
| Week 2  | Chemical Kinetics  |
| 22 <sup>nd</sup> January – 27 <sup>th</sup>       | Mathematical characteristics of simple chemical reactions, molecularity and  |
| January   | order of reaction. Zero order ,Ist order Second order, third order           |
|   | reactions and their mathematical derivations for their rate constants.and    |
|   | their mathematical derivations or their rate constants.                      |
|   | their mathematical activations of their rate constants.                      |
| 28 <sup>th</sup> January                          | SUNDAY   |
| Week 3  | Chemical Kinetics  |
| 29 <sup>th</sup> January - 03 <sup>th</sup>       | Half life period, average life period, determination of order reaction.      |
| February  | Differential method, method of integration. Method of half life period and   |
|   | isolation method. Pseudo unimolecular reactions                              |
| 04 <sup>th</sup> February                         | SUNDAY   |
| Week 4  | Chemical Kinetics  |
| 05 <sup>th</sup> February 10 <sup>th</sup>        | Revision, Assignment and Test  |
| February  | Action, Assignment and Test  |
| 11 <sup>th</sup> February                         | SUNDAY   |
| Week 5  | Electrochemistry-I   |
| 12 <sup>th</sup> February – 17t <sup>h</sup>      | Electrical transport conduction in metal and in electrolyte solutions,       |
| February  | specific conductance and equivalent conductance. Measurement of equivalent   |
|   | conductance. Variation of equivalent conductance and specfic conductance     |
|   | with dilution  |
| 18 <sup>th</sup> February                         | SUNDAY   |
| Week 6  | Electrochemistry-I   |
| 19 <sup>th</sup> February – 24 <sup>th</sup>      | migration of ions, Kohlrausch's law, Arhenious theory of electrolyte         |
| February  | dissolution and its limitations. Weak and strong electrolytes. Ostwald's     |
|   | dilution law and its uses and limitation.                                    |
| 25th Eabannana                                    |  |
| 25 <sup>th</sup> February<br>Week 7               | SUNDAY  Floatrochomictry-I   |
| 26thFebruary-02nd                                 | Electrochemistry-I   |
| March   | Revision, Assignment and Test  |
| 03 <sup>th</sup> March                            | SUNDAY   |
| Week 8  | Electrochemistry-II  |
| 04th March-09th                                   | Debye-Huckel onsager equation for strong electrolytes (elementary treatment  |
| or march o  |  |
| or march os                                       | only), transport number and its determination by Hittorf method              |

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2023-24 Semester: Even

Name of Assistant Prof./Associate Prof./Professor: Dr. Ravinder Singh

Class: B.Sc. 2<sup>nd</sup> /4<sup>th</sup> /6<sup>th</sup> Semester (Chem. Hons.)

| 10 <sup>th</sup> March                          | SUNDAY  |
|---|---|
| Week 9  | Electrochemistry-II   |
| 11 <sup>th</sup> March–16 <sup>th</sup> March   | Transport number and its determination by moving boundary                 |
|   | method. Application of conductivity measurements, determination of        |
|   | solubility product of spraningly soluble salts.                           |
| 17 <sup>th</sup> March                          | SUNDAY  |
| Week 10   | Electrochemistry-II   |
| 18 <sup>th</sup> March— 22 <sup>th</sup>        | Determination of degree of dissolution , Ka for weak acids.               |
| March   |   |
| 24 <sup>th</sup> March                          | SUNDAY  |
| 23 <sup>rd</sup> March - 31 <sup>st</sup>       | Holidays  |
| March 31st March                                | SUNDAY  |
| Week 12   |   |
| 01st April – 06th April                         | Electrochemistry-II   |
| 07 <sup>th</sup> April                          | Revision, Assignment and Test SUNDAY                                      |
| Week 13   | Theromchemistry and chemical energaties:                                  |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April  |   |
|   |   |
| 1.4th A1  | during chemical reactions Derivation of Ist law of thermodynamus.  SUNDAY |
| 14 <sup>th</sup> April<br>Week 14               |   |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April | Theromchemistry and chemical energaties:                                  |
|   | Heat of reaction, enthalpy and enthalpy change. Enthalpy of formation,    |
|   | combustion, neutralisation, solution, 16 vaporisation, sublimation        |
|   | hydration and fusion, calorfic value of foods. Bond energy and its        |
|   | calculation. Hess's law of heat summation and its application for the     |
|   | calculation of various enthalpies of reaction. Kirchhoff's equation,      |
| 21stApril                                       | SUNDAY  |
| Week 15   | Theromchemistry and chemical energaties:                                  |
| 22 <sup>nd</sup> April – 27 April               | Spontaneous processes. Criteria of spontaneity., entropy and free energy. |
|   | Why crisis of energy if conserved in nature.                              |
| 28 <sup>th</sup> April                          | SUNDAY  |
| Week 16   | Theromchemistry and chemical energaties:                                  |
| 29 <sup>th</sup> April-30 <sup>th</sup> April   | Revision, Assignment and Test   |

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Name of Assistant Prof./Associate Prof./Professor: Dr. Ravinder Singh

Class: B.Sc. 2<sup>nd</sup> /4<sup>th</sup> /6<sup>th</sup> Semester (Chem. Hons.)

| 16th January 2023 to 16th   | May 2023 [B.Sc. 4th Semester Chemistry Hons.] Physical Chemistry  |
|---|---|
| Week 1  | Thermodynamics - II   |
| 16 <sup>th</sup> January – 20 <sup>th</sup>   | Second law of thermodynamics. Need of the law, different definitions of the   |
| January   | law, Carnot Cycle and its efficiency. Carnot theorem. Thermodynamic scale   |
|   | of temperature.   |
| 21st January  | SUNDAY  |
| Week 2  | Thermodynamics - II   |
| 22 <sup>nd</sup> January – 27 <sup>th</sup>   | Concept of entropy, entropy as a state function of V and T, entropy as a  |
| January   | function of P and T. Entropy change in physical processes. Clausius   |
|   | inequility. Entropy as criteria of spontaneity and equilibrium. Entropy   |
|   | change in ideal gases and mixing of gases,  |
| 28 <sup>th</sup> January  | SUNDAY  |
| Week 3  | Thermodynamics - II   |
| 29 <sup>th</sup> January - 03 <sup>th</sup>   | work function, Gibb, s free energy function. Gibbs function (G) and   |
| February  | Helmholtz function (A) as thermodynamic function. Criteria of spontaneity   |
|   | of reversible processes in terms of enthalpy change, entropy change, work   |
|   | function and free energy function.  |
| 04 <sup>th</sup> February   | SUNDAY  |
| Week 4  | Thermodynamics - II   |
| 05 <sup>th</sup> February 10 <sup>th</sup>  | Revision, Assignment and Test   |
| February  |   |
|   |   |
| 11 <sup>th</sup> February   | SUNDAY  |
| Week 5  | Thermodynamics - III  |
| Week 5<br>12 <sup>th</sup> February – 17t <sup>h</sup>  | Thermodynamics - III<br>Variation of G and A with P,V and T.Gibb Helmholtz equation and its   |
| Week 5<br>12 <sup>th</sup> February – 17t <sup>h</sup><br>February  | Thermodynamics - III<br>Variation of G and A with P,V and T.Gibb Helmholtz equation and its<br>application, clausius-clapeyron equation   |
| Week 5<br>12 <sup>th</sup> February – 17t <sup>h</sup><br>February<br>18 <sup>th</sup> February   | Thermodynamics - III Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius- clapeyron equation  SUNDAY  |
| Week 5 12 <sup>th</sup> February – 17t <sup>h</sup> February  18 <sup>th</sup> February Week 6  | Thermodynamics - III Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius— clapeyron equation  SUNDAY Thermodynamics - III   |
| Week 5 12 <sup>th</sup> February – 17t <sup>h</sup> February  18 <sup>th</sup> February  Week 6 19 <sup>th</sup> February – 24 <sup>th</sup>  | Thermodynamics - III Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius- clapeyron equation  SUNDAY  |
| Week 5 12 <sup>th</sup> February – 17t <sup>h</sup> February  18 <sup>th</sup> February Week 6  | Thermodynamics - III Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius— clapeyron equation  SUNDAY Thermodynamics - III   |
| Week 5 12 <sup>th</sup> February – 17t <sup>h</sup> February  18 <sup>th</sup> February  Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February   | Thermodynamics - III Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius— clapeyron equation  SUNDAY Thermodynamics - III Nernst heat theorem. Third law of thermodynamics and its applications.  |
| Week 5 12 <sup>th</sup> February – 17th February  18 <sup>th</sup> February Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February 25 <sup>th</sup> February Week 7 26thFebruary–02 <sup>nd</sup>   | Thermodynamics - III Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius- clapeyron equation  SUNDAY Thermodynamics - III Nernst heat theorem. Third law of thermodynamics and its applications.  SUNDAY Thermodynamics - III   |
| Week 5 12 <sup>th</sup> February – 17t <sup>h</sup> February  18 <sup>th</sup> February  Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February 25 <sup>th</sup> February  Week 7   | Thermodynamics - III Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius— clapeyron equation  SUNDAY Thermodynamics - III Nernst heat theorem. Third law of thermodynamics and its applications.  SUNDAY Thermodynamics - III Partial molar quantities. Chemical potential. Gibb's Duhem equation. Gibb, s  |
| Week 5 12 <sup>th</sup> February – 17th February  18 <sup>th</sup> February Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February 25 <sup>th</sup> February Week 7 26thFebruary–02 <sup>nd</sup>   | Thermodynamics - III Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius— clapeyron equation  SUNDAY Thermodynamics - III Nernst heat theorem. Third law of thermodynamics and its applications.  SUNDAY Thermodynamics - III Partial molar quantities. Chemical potential. Gibb's Duhem equation. Gibb, s adsorption equation and its application, variation of chemical potential   |
| Week 5 12 <sup>th</sup> February – 17th February  18 <sup>th</sup> February Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February 25 <sup>th</sup> February Week 7 26thFebruary–02 <sup>nd</sup>   | Thermodynamics - III Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius— clapeyron equation  SUNDAY Thermodynamics - III Nernst heat theorem. Third law of thermodynamics and its applications.  SUNDAY Thermodynamics - III Partial molar quantities. Chemical potential. Gibb's Duhem equation. Gibb, s  |
| Week 5 12 <sup>th</sup> February – 17th February  18 <sup>th</sup> February Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February 25 <sup>th</sup> February Week 7 26thFebruary–02 <sup>nd</sup> March  03 <sup>th</sup> March Week 8  | Thermodynamics - III  Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius— clapeyron equation  SUNDAY  Thermodynamics - III  Nernst heat theorem. Third law of thermodynamics and its applications.  SUNDAY  Thermodynamics - III  Partial molar quantities. Chemical potential. Gibb's Duhem equation. Gibb, s adsorption equation and its application, variation of chemical potential with temperature and pressure  |
| Week 5 12 <sup>th</sup> February – 17th February  18 <sup>th</sup> February Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February 25 <sup>th</sup> February Week 7 26thFebruary–02 <sup>nd</sup> March   | Thermodynamics - III Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius— clapeyron equation  SUNDAY Thermodynamics - III Nernst heat theorem. Third law of thermodynamics and its applications.  SUNDAY Thermodynamics - III Partial molar quantities. Chemical potential. Gibb's Duhem equation. Gibb, s adsorption equation and its application, variation of chemical potential with temperature and pressure  SUNDAY   |
| Week 5 12 <sup>th</sup> February – 17th February  18 <sup>th</sup> February Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February 25 <sup>th</sup> February Week 7 26thFebruary–02 <sup>nd</sup> March  03 <sup>th</sup> March Week 8 04 <sup>th</sup> March–09 <sup>th</sup> 10 <sup>th</sup> March | Thermodynamics - III  Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius- clapeyron equation  SUNDAY  Thermodynamics - III  Nernst heat theorem. Third law of thermodynamics and its applications.  SUNDAY  Thermodynamics - III  Partial molar quantities. Chemical potential. Gibb's Duhem equation. Gibb, s adsorption equation and its application, variation of chemical potential with temperature and pressure  SUNDAY  Thermodynamics - III  Partial molar quantities. Themselves application, variation of chemical potential with temperature and pressure  SUNDAY  Thermodynamics - III |
| Week 5 12 <sup>th</sup> February – 17th February  18 <sup>th</sup> February Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February 25 <sup>th</sup> February Week 7 26thFebruary–02 <sup>nd</sup> March  03 <sup>th</sup> March Week 8 04 <sup>th</sup> March–09 <sup>th</sup>                        | Thermodynamics - III Variation of G and A with P,V and T.Gibb Helmholtz equation and its application, clausius— clapeyron equation  SUNDAY Thermodynamics - III Nernst heat theorem. Third law of thermodynamics and its applications.  SUNDAY Thermodynamics - III Partial molar quantities. Chemical potential. Gibb's Duhem equation. Gibb, s adsorption equation and its application, variation of chemical potential with temperature and pressure  SUNDAY Thermodynamics - III Revision, Assignment and test  |

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Class: B.Sc. 2<sup>nd</sup> /4<sup>th</sup> /6<sup>th</sup> Semester (Chem. Hons.)

|  | irreversible cells reversible electrodes, types of reversible electrodes,   |
|--|---|
|  |   |
|  | metal electrodes, gas metal electrode, metal insoluble salt on ions and     |
| 4 =th > 5  | redox electrodes, electrode reactions,                                      |
| 17 <sup>th</sup> March                                     | SUNDAY  |
| Week 10<br>18 <sup>th</sup> March– 22 <sup>th</sup>        | Electrochemistry-II   |
| March  | cell voltage ,function of salt bridge, electrode potential and its          |
| IVIUI CII  | determination. Standard hydrogen electrode, reference electrode, standard   |
|  | cell, sign convention. Electrochemical series and its significance.         |
| 24 <sup>th</sup> March                                     | SUNDAY  |
| 23 <sup>rd</sup> March - 31 <sup>st</sup>                  | Holidays  |
| March  |   |
| 31st March   | SUNDAY  |
| Week 12<br>01 <sup>st</sup> April – 06 <sup>th</sup> April | Electrochemistry-II   |
| or April - 00 April  | Nernst equation for a reversible electrode and cell. Calculation of         |
|  | thermodynamic quantities of a cell reaction • G, • H and K. Polarisation    |
|  | over potential and hydrogen over voltage                                    |
| 07 <sup>th</sup> April                                     | SUNDAY  |
| Week 13  | Electrochemistry-III  |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April             | Definition of pH. Determination of pH using hydrogen, quinhydron and glass  |
|  | electrode by potentiometric method. Buffers solution, Buffer action,        |
|  | Henderson - Hazel equation. Hydrolysis of salts, corrosion, types, theories |
|  | and methods of controlling it.  |
| 14 <sup>th</sup> April                                     | SUNDAY  |
| Week 14  | Chemical Kinetics   |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April            | Experimental methods of chemical kinetics: conductometric, potentiometric   |
|  | , optical method , polarimetery and spectrophotometer.                      |
| 21stApril  | SUNDAY  |
| Week 15  | Chemical Kinetics   |
| 22 <sup>nd</sup> April – 27 April                          | Theories of reaction rates, effect of temperature on rate of reaction.      |
|  | Simple collision theory based upon transition state, hard sphere model      |
|  |   |
|  | theory (equilibrium hypothesis). Expression for the rate constants based on |
| 20th A 1   | equilibrium constant their thermodynamic aspect                             |
| 28 <sup>th</sup> April<br>Week 16                          | SUNDAY  |
|  | Chemical Kinetics   |
| 29 <sup>th</sup> April-30 <sup>th</sup> April              | Revision, Assignment and test   |

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Class: B.Sc. 2<sup>nd</sup> /4<sup>th</sup> /6<sup>th</sup> Semester (Chem. Hons.)

| 16th January 2024 to 30th                    | April 2024 [B.Sc. 6 <sup>th</sup> Semester Chemistry Hons.] Organic Chemistry-I                |
|--|--|
| Week 1                                       | <b>Topics in Biological Chemistry:</b> Introduction to enzymes, nomenclature, characteristics. |
| 16 <sup>th</sup> January – 20 <sup>th</sup>  | Practicals:  |
| January                                      |  |
| 21st January                                 | SUNDAY   |
| Week 2                                       | Topics in Biological Chemistry: General picture of mechanism of enzymes action, co-            |
| 22 <sup>nd</sup> January – 27 <sup>th</sup>  | enzymes: co-enzymes derived from niacin and thiamine, lipoic acid, co-enzyme- A.               |
| January                                      | Practicals:  |
| 28th January                                 | SUNDAY   |
| Week 3                                       | Topics in Biological Chemistry: Energy production in biological system, glycolysis,            |
| 29 <sup>th</sup> January - 03 <sup>th</sup>  | tricarboxylic acid cycle   |
| February                                     | Practicals:  |
| 04 <sup>th</sup> February                    | SUNDAY   |
| Week 4                                       | Revision, Assignment and test  |
| 05 <sup>th</sup> February 10 <sup>th</sup>   | Practicals:  |
| February                                     |  |
| 11 <sup>th</sup> February                    | SUNDAY   |
| Week 5                                       | Fermentation   |
| 12 <sup>th</sup> February – 17t <sup>h</sup> | Anaerobic and aerobic fermentation, production of alcohol, citric acid and lactic acid         |
| February                                     | Practicals:  |
| 18 <sup>th</sup> February                    | SUNDAY   |
| Week 6                                       | Fats, Oil and Detergents:  |
| 19 <sup>th</sup> February – 24 <sup>th</sup> | Occurrence, chemical composition and importance, hydrogenated oils, Rancidity, acid            |
| February                                     | value, sponification and iodine numbers.   |
|  | Practicals:  |
| 25 <sup>th</sup> February                    | SUNDAY   |
| Week 7                                       | Fats, Oil and Detergents:  |
| 26thFebruary-02nd                            | Difference between toilet and washing soaps, comparison o f soap and detergents,               |
| March  | classification and principle of cleansing action of detergents.                                |
|  | Practicals:  |
| 03 <sup>th</sup> March                       | SUNDAY   |
| Week 8                                       | Revision, Assignment and test  |
| 04 <sup>th</sup> March-09 <sup>th</sup>      | Practicals:  |
| 10 <sup>th</sup> March                       | SUNDAY   |
| Week 9                                       | Drugs-I  |
| 11th March-16th March                        | Introduction, relation of chemical structure and physiological activity with suitable          |
|  | examples.  |
|  | Practicals:  |
| 17 <sup>th</sup> March                       | SUNDAY   |
| 17 Mai Cii                                   | DUMPHI   |

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2023-24 Semester: Even

Name of Assistant Prof./Associate Prof./Professor: Dr. Ravinder Singh

Class: B.Sc. 2<sup>nd</sup> /4<sup>th</sup> /6<sup>th</sup> Semester (Chem. Hons.)

| XX 1 10   |   |
|---|---|
| Week 10<br>  18 <sup>th</sup>   March-   22 <sup>th</sup> | Drugs-I   |
|   | Mechanism of chemotherapeutic action. Nomenclature of organic chemical systems, |
| March   | stereochemical notations.   |
|   | General aspects, preparation and uses of the following drugs:                   |
|   | (i) Analgesics and antipyretics : paracetamol, Aspirin                          |
|   | (i) Anti-inflammatory: Ibuprofen  |
|   | (iii) Antiseptics and disinfectants Chloro cresol, povidone – Iodine            |
|   | Practicals:   |
| 24th March  | SUNDAY  |
| 23 <sup>rd</sup> March - 31 <sup>st</sup>                 | Holidays  |
| March   |   |
| 31st March  | Sunday  |
| Week 12   | Drugs-II  |
| 01st April – 06th April                                   | General aspects, preparation and uses of the following drugs:                   |
|   | (i) Sulpha: Sulphacetamide  |
|   | (ii) Local anaesthetics : Benzocaine  |
|   | Practicals:   |
| 07 <sup>th</sup> April                                    | SUNDAY  |
| Week 13   |   |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April            | Drugs-II  |
| 8 April – 13 April  | General aspects, preparation and uses of the following drugs:                   |
|   | (iii) Anti amoebic : Metronidazole  |
|   | (iv) Antimalarials : Chloroquinine  |
|   | Practicals:   |
| 14 <sup>th</sup> April                                    | SUNDAY  |
| Week 14   | Revision, Assignment and test   |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April           | Practicals:   |
| 21stApril   | SUNDAY  |
| Week 15   | Drugs-II  |
| 22 <sup>nd</sup> April – 27 April                         | General aspects, preparation and uses of the following drugs:                   |
|   | (v) Antihistamines : Chlorophenizamine Maleate                                  |
|   | (vi) Antifungal Undecylenic acid  |
|   | Practicals:   |
| 28 <sup>th</sup> April                                    | SUNDAY  |
| Week 16   | Drugs-II  |
| 29th April-30th April                                     | General aspects, preparation and uses of the following drugs:                   |
| •   | (vii) Insect repellants: Dibutyl phthalate                                      |
|   | (vii) Antibiotics Chloroamphenicol  |
|   | Practicals:   |
|   | 1 facticals.  |

|   | Summary of Lesson Plan of College Faculty  |
|---|--|
| Name of College: Pt. Nel<br>Semester: Even          | ki Ram Sharma Government College, Rohtak Academic Session 2023-24                      |
| Name of Asstt./Ass. Prol                            | F. – Dr. Rinki   |
| Class: B.Sc Chemistry Hons. 2 <sup>nd</sup> SEM     |  |
| Paper- XIV (Theory) Or                              | rganic Chemistry   |
|   |  |
| 1 <sup>st</sup> January – 15 <sup>t</sup> h         | Practical Slot   |
| January   | Alkenes  |
|   | Nomenclature of alkenes, mechanisms of dehydration of alcohols and dehydrohald         |
| Week 1  | of alkyl halides, Regioselectivity in alcohol dehydration. The Saytzeff rule, Hofma    |
| 16 January - 20                                     | elimination, physical p roperties and relative stabilities of alkenes                  |
| January   |  |
| 21st January  | SUNDAY   |
|   | Alkenes  |
| Week 2  | Chemical reactions of alkenes, mechanisms involved in hydrogenation, electrophil       |
|   | free radical additions, Markownikoff's rule, hydroboration–oxidation, oxymercura       |
| 22 <sup>rd</sup> January – 27 <sup>th</sup> January | reduction, Epoxidation, ozonolysis, hydration, hydroxylation and oxidation with K      |
| Januar y  | polymerization of alkenes, substitution at the allylic and vinylic positions of alkene |
|   | Industrial applications of ethylene and propene  |
| 28th January  | SUNDAY   |
| 20 January  | SUNDAI   |
| Week 3  | Coal, petroleum and petrochemicals:  |
| 29th January - 03th                                 | Coal tar distillation and coal tar chemicals, petroleum origin, fractionation          |
| February  | cracking, reforming and aromatisation, petrochemicals, synthetic fuels, octane and     |
|   | cetane numbers, antiknock additives  |
|   |  |
|   |  |
| 04 <sup>th</sup> February                           | SUNDAY   |
| Week 4  | Arenes and Aromaticity   |
| 05 <sup>th</sup> February 10 <sup>th</sup>          | Nomenclature of benzene derivatives. The Aryl group. Aromatic nucleus and side         |
| February  | chain. Structure of benzene: molecular formula and Kekule structure. Stability         |
|   | and carbon-carbon bond lengths of benzene, resonance structure, MO picture.            |
|   | ı  |

| 11 <sup>th</sup> February                                | SUNDAY   |
|--|--|
| Week 5   |  |
| 12 <sup>th</sup> February – 17t <sup>h</sup><br>February | Arenes and Aromaticity   |
| rebluary   | Aromaticity: the Huckel rule, aromatic ions, annulenes up to 10 carbon atoms, aro anti - aromatic and non – aromatic compounds. Aromatic electrophilic substitution pattern of the mechanism, role of sigma and pia-complexes, mechanism of nitratic halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams.         |
| 18th February  | SUNDAY   |
| Week 6   | Arenes and Aromaticity   |
| 19 <sup>th</sup> February – 24 <sup>th</sup> February    | Activating, deactivating substituents and orientation and *ortho/para ratio. *Side chain reactions of benzene derivatives, *Birch reduction. * Methods of formation and chemical reactions of alkylbenzenes, alkynylbenzenes and biphenyl  |
| 25 <sup>th</sup> February                                | SUNDAY   |
| Week 7   | Dienes and Alkynes   |
| 26thFebruary-02nd<br>March                               | Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of Allenes and butadiene, Methods of formation, polymerization. Chemical reactions: 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction, Nomenclature, structure and bonding in alkynes. Methods of formation |
| 03 <sup>th</sup> March                                   | SUNDAY   |
| Week 8   | Dienes and Alkynes   |
| 04 <sup>th</sup> March-09 <sup>th</sup>                  | Chemical reactions of alkynes, acidity of alkynes. Mechanism of electrophilic and nucleophilic addition reactions, hydroboration-oxidation of alkynes, metal-ammonia reductions, oxidation and polymerization.  Cycloalkenes: Methods of formation, conformation and chemical reactions of   |
|  | cycloalkenes   |
| 10 <sup>th</sup> March                                   | SUNDAY   |

| Week 9   | Poly Nuclear Hydrocarbons  |
|--|--|
| 11 <sup>th</sup> March–16 <sup>th</sup><br>March             | Haworth synthesis of naphthalene and phenanthene, pschorr synthesis of phenanthrene, synthesis of anthracene involving Friedal crafts acylation of benzene with phthalic anhydride and Diels Alder reaction between 1,3-butadiene and 1,4-naphthaquinone, reaction of naphthalene, anthracene and phenanthrene, relative reactivities at different positions and mechanism of electrophilic substitution reactions in naphthalene, anthracene, and phenanthrene. |
| 17 <sup>th</sup> March                                       | SUNDAY   |
| Week 10<br>18 <sup>th</sup> March– 22 <sup>th</sup><br>March | Nomenclature and classes of alkyl halides, methods of formation, chemical reactions. Mechanisms and stereochemistry of nucleophilic substitution reactions of alkyl halides,   |
| 24 March   | SUNDAY Week 11 23-31 March Holi Break  |
| Week 12  | Alkyl and Aryl Halides   |
| 01 <sup>st</sup> April – 06 <sup>th</sup><br>April           | Methods of formation, Reactions of aryl halides  |
| 07 <sup>th</sup> April                                       | SUNDAY   |
| Week 13  | Alkyl and Aryl Halides   |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April               | SN2 and SN1 reactions with energy profile diagrams. Study of elimination reactions in alkyl halides, E1. And E2 mechanism, substitution vs. elimination, factors affecting substitution/elimination.   |
| 14 <sup>th</sup> April                                       | SUNDAY   |
| Week 14  15 <sup>th</sup> April – 20 <sup>th</sup> April     | The addition elimination and the elimination-addition mechanisms of nucleophilic aromatic substitution reactions. Relative reactivities of alkyl halides vs allyl, vinyl and aryl halides.   |
| 21stApril  | SUNDAY   |
| Week 15 22 <sup>nd</sup> April – 27 April                    | <b>Polyhalogen compounds</b> : chloroform, carbon tetrachloride. Synthesis and uses of DDT and BHC   |
| 28 <sup>th</sup> April                                       | SUNDAY   |

| Week 16                                       | Brief Section A, B, C, D Queries/ Doubts |
|---|--|
| 29 <sup>th</sup> April-30 <sup>th</sup> April |  |

| Name of College: Pt. Neki Ram Sharma Government College, Rohtak    |  |  |
|--|--|--|
| Academic Session 2023-24   |  |  |
| Name of Asstt./Ass. P  | Name of Asstt./Ass. Prof. – Dr. ANJU SIWACH  |  |
| Class: B.Sc. (Hons.) Chemistry 2 <sup>nd</sup> SEMESTER            |  |  |
| Name of Subject: INO   | Name of Subject: INORGANIC CHEMISTRY   |  |
| 1 <sup>st</sup> January – 15 <sup>th</sup><br>January              | Practical Slot   |  |
| Week 1<br>16 January - 20  | s-Block Elements: Comparative study, diagonal relationships, salient features of hydrides,                         |  |
| January  |  |  |
| 21st January   | SUNDAY   |  |
| Week 2<br>22 <sup>rd</sup> January –<br>27 <sup>th</sup> January   | solvation and complexation tendencies including their function in biosystems, an introduction to alkyls and aryls. |  |
| 28 <sup>th</sup> January   | SUNDAY   |  |
| Week 3 29 <sup>th</sup> January - 03 <sup>th</sup> February        | Theory of Precipitation: theory, purification of precipitates, co-precipitation and post-precipitation             |  |
| 04 <sup>th</sup> February  | SUNDAY   |  |
| Week 4<br>05 <sup>th</sup> February 10 <sup>th</sup><br>February   | Theory of Qualitative and Quantitative Inorganic Analysis:   |  |
| 11 <sup>th</sup> February  | SUNDAY   |  |
| Week 5<br>12 <sup>th</sup> February –<br>17t <sup>h</sup> February | Chemistry of analysis of various groups of basic and acidic radicals,  |  |
| 18 <sup>th</sup> February  | SUNDAY   |  |
| Week 6<br>19 <sup>th</sup> February –<br>24 <sup>th</sup> February | chemistry of identification of acid radicals in typical combinations.  |  |
| <b>√</b>   |  |  |

| 25th Eabannana                           | SUNDAY  |
|--|---|
| 25 <sup>th</sup> February<br>Week 7      | Chemistry of interferences of acid radicals including their removal in the analysis of basic radicals.  |
| 26th February – 02nd                     | Chemistry of interferences of acid fadicals including their removal in the analysis of basic fadicals.  |
| March                                    |   |
| 03 <sup>th</sup> March                   | SUNDAY  |
| Week 8                                   | Revision, Assignment and Test   |
| 04 <sup>th</sup> March-09 <sup>th</sup>  | , 0   |
| 10 <sup>th</sup> March                   | SUNDAY  |
| Week 9                                   | <b>p-Block Elements - I</b> Comparative study (including diagonal relationship) of groups 13-17         |
| 11 <sup>th</sup> March–16 <sup>th</sup>  | elements, compounds like hydrides, oxides of groups 13-16   |
| March                                    |   |
| 17 <sup>th</sup> March                   | SUNDAY  |
| Week 10                                  | oxyacids and halides of groups 13-16,   |
| 18 <sup>th</sup> March– 22 <sup>th</sup> | hydrides of boron - diborane and higher boranes, borazine, borohydrides. (Unit Test)                    |
| March                                    |   |
| 24 March                                 | SUNDAY  |
| Week 11                                  | 23-31 March Holi Break  |
| Week 12                                  | <b>p-Block Elements - II</b> Chemistry of fullerenes, carbides, fluorocarbons, silicates (structural    |
| 01st April - 06th                        | principle) tetrasulphur tetranitride,   |
| April                                    | Basic properties of halogens, interhalogens and polyhalides. <b>Chemistry of Noble Gases</b> : Chemical |
|  | properties of the noble gases,  |
|  |   |
| 07 <sup>th</sup> April                   | SUNDAY  |
| Week 13                                  | Chemistry of xenon, structure and bonding in xenon compounds. ( <b>Unit Test</b> )                      |
| 8 <sup>th</sup> April – 13 <sup>th</sup> |   |
| April                                    |   |
| 14 <sup>th</sup> April                   | SUNDAY  |
| Week 14                                  | Assignment+Viva   |
| 15 <sup>th</sup> April –                 | Revision  |
| 20 <sup>th</sup> April                   | CITINID ANZ   |
| 21 <sup>st</sup> April                   | SUNDAY  |
|  |   |
| Week 15                                  | Brief Section A, B, C, D Queries/ Doubts  |
| 22 <sup>nd</sup> April – 27              |   |
| April                                    | CITINID ANZ   |
| 28 <sup>th</sup> April                   | SUNDAY  |
| Week 16                                  | Brief Section A, B, C, D Queries/ Doubts  |
| 29 <sup>th</sup> April-30 <sup>th</sup>  | Division in Di Querron Dounes   |
| April                                    |   |
|  |   |

| Name of College: Pt. Neki Ram Sharma Government College, Rohtak<br>Academic Session 2023-24 |   |  |
|---|---|--|
| Name of Asstt./Ass. Prof. – Dr. ANJU SIWACH   |   |  |
| Class: B.Sc. (Hons.) Ch   | emistry 4 <sup>nd</sup> SEMESTER  |  |
| Name of Subject: INO  | RGANIC CHEMISTRY  |  |
| 1 <sup>st</sup> January – 15 <sup>th</sup><br>January                                       | Practical Slot  |  |
| Week 1<br>16 January - 20<br>January  | Chemistry of Elements of Second and Third Transition Series General characteristics, comparative treatment with their 3d-analogues in respect of ionic radii, |  |
| 21st January  | SUNDAY  |  |
| Week 2<br>22 <sup>rd</sup> January –<br>27 <sup>th</sup> January                            | oxidation states, magnetic behaviour, spectral properties and stereochemistry.  |  |
| 28 <sup>th</sup> January  | SUNDAY  |  |
| Week 3<br>29 <sup>th</sup> January - 03 <sup>th</sup><br>February                           | Chemistry of Mo and W in different oxidation states (Unit Test)   |  |
| 04 <sup>th</sup> February   | SUNDAY  |  |
| Week 4<br>05 <sup>th</sup> February 10 <sup>th</sup><br>February                            | Isopolyacids of Mo and W: aqueous chemistry of Mo and W(VI),  |  |
| 11 <sup>th</sup> February   | SUNDAY  |  |
| Week 5<br>12 <sup>th</sup> February –<br>17t <sup>h</sup> February                          | isopolymolybedates and isopolytungustates   |  |
| 18 <sup>th</sup> February   | SUNDAY  |  |
| Week 6<br>19 <sup>th</sup> February –   |   |  |

| 24 <sup>th</sup> February                 | Acids and Bases:   |
|---|--|
| 25 <sup>th</sup> February                 | SUNDAY   |
| Week 7                                    | Arrhenius, Bronsted- Lowry, the Lux- Flood, solvent system and Lewis concepts of acids and   |
|   | bases  |
| 26thFebruary-02nd<br>March                | ouses -  |
| 03 <sup>th</sup> March                    | SUNDAY   |
| Week 8                                    |  |
| 04 <sup>th</sup> March-09 <sup>th</sup>   | Revision, Assignment and Test  |
| 10 <sup>th</sup> March                    | SUNDAY   |
| Week 9                                    | General Principles of Metallurgy:  |
| 11 <sup>th</sup> March–16 <sup>th</sup>   | General principles of metallurgy, occurrence of metals with special emphasis on mineral wealth of  |
|   | India, calcination roasting, smelting, bessemerization,  |
| March                                     | mora, caremation rousting, smerting, cossementation,   |
| 17 <sup>th</sup> March                    | SUNDAY   |
| Week 10                                   | various methods of concentration, purification and refining (such as parting process, zone refining,   |
| 18th March- 22th                          | oxidation refining, electrolytic refining and solvent extraction   |
| March                                     |  |
| 24 March                                  | SUNDAY   |
| Week 11                                   | 23-31 March Holi Break   |
| Week 12                                   | metallurgy of important metals like Ag, Au, Zn, Cu, Ni. (Unit Test)  |
| 01 <sup>st</sup> April - 06 <sup>th</sup> |  |
| April                                     |  |
| 07 <sup>th</sup> April                    | SUNDAY   |
| Week 13                                   | Chemistry of Lanthanide Elements:  |
| 8 <sup>th</sup> April – 13 <sup>th</sup>  | Electronic structure, oxidation states and ionic radii and lanthanide contraction, complex   |
| April                                     | formation, occurrence and isolation, lanthanide compounds.   |
| a 4th a an                                |  |
| 14 <sup>th</sup> April                    | SUNDAY   |
| Week 14                                   | Chemistry of Actinides:  Constal features and shamistry of actinides, shamistry of constaling of Nn. By and Am from H.   |
| 15 <sup>th</sup> April –                  | General features and chemistry of actinides, chemistry of separation of Np, Pu and Am from U, similarities between the later actinides and the later lanthanides. ( <b>Unit Test</b> ) |
| 20 <sup>th</sup> April                    | , , , , ,  |
| 21stApril                                 | SUNDAY   |
|   |  |
| Week 15                                   | Brief Section A, B, C, D Queries/ Doubts   |
| 22 <sup>nd</sup> April – 27               |  |
| April                                     |  |
| 28 <sup>th</sup> April                    | SUNDAY   |
|   |  |
| Week 16                                   | Brief Section A, B, C, D Queries/ Doubts   |
| 29 <sup>th</sup> April-30 <sup>th</sup>   |  |
| April                                     |  |
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| Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2023-24 Semester: |   |  |  |
|--|---|--|--|
| Even Name of Asstt./Ass. Prof. – Dr. Suman Bhatti  |   |  |  |
| Class: B.Sc Chemistry Hons. 4 <sup>th</sup> SEM  |   |  |  |
| Name of Subject: Org   |   |  |  |
| , c  | •   |  |  |
| 1 <sup>st</sup> January – 15 <sup>th</sup>   | Practical Slot  |  |  |
| January  |   |  |  |
| Week 1   |   |  |  |
| 16 January - 20  | Infrared (IR) absorption spectroscopy   |  |  |
| January  | Molecular vibrations, Hooke's law, selection rules, intensity and position of IR bands,   |  |  |
| 21st Ionwowy   | measurement of IR spectrum, fingerprint region  |  |  |
| 21st January   | SUNDAY  |  |  |
|  |   |  |  |
| Week 2   | Characteristic absorptions of various functional groups and interpretation of IR spectra of simple  |  |  |
| 22 <sup>rd</sup> January –   | organic compounds. *Hydrocarbons (saturated and unsaturated), hydroxy compounds, aldehydes, ketones, esters, anhydrides, amides, amines and nitrocompounds.                                       |  |  |
| 27 <sup>th</sup> January   | ketones, esters, anniyarides, annues, annues and introcompounds.  |  |  |
|  |   |  |  |
| 28 <sup>th</sup> January   | SUNDAY  |  |  |
| Week 3   |   |  |  |
| 29 <sup>th</sup> January - 03 <sup>th</sup>  | Applications of IR spectroscopy in structure elucidation of simple organic compounds.   |  |  |
| February   |   |  |  |
|  |   |  |  |
| 04 <sup>th</sup> February  | SUNDAY  |  |  |
| Week 4   |   |  |  |
| 05 <sup>th</sup> February 10 <sup>th</sup>   | Aldehydes and Ketones   |  |  |
| February   | Nomenclature and structure of the carbonyl group. Synthesis of aldehydes and ketones with   |  |  |
|  | particular reference to the synthesis of aldehydes from acid chlorides, advantage of oxidation of alcohols with chromium trioxide (Sarett reagent) pyridinium chlorochromate (PCC) and pyridinium |  |  |
|  | dichromate., * synthesis of aldehydes and ketones using 1,3-dithianes,  |  |  |
| 11 <sup>th</sup> February  | SUNDAY  |  |  |
| Week 5   |   |  |  |
| 12 <sup>th</sup> February –  | *Gatterman aldehyde synthesis, *Gatterman Koch reaction, *synthesis of ketones from nitriles and  |  |  |
| 17th February  | from carboxylic acids. Physical properties. Comparison of reactivities of aldehydes and ketones.  |  |  |
| 18 <sup>th</sup> February  | SUNDAY  |  |  |
| Week 6   |   |  |  |
| 19 <sup>th</sup> February –  | Mechanism of nucleophilic additions to carbonyl group with particular emphasis on benzoin, aldol,   |  |  |
| = = = = = J  |   |  |  |

| 24 <sup>th</sup> February                | Perkin and Knoevenagel condensations. Condensation with ammonia and its derivatives.  |
|--|---|
|  |   |
| 25 <sup>th</sup> February                | SUNDAY  |
| Week 7                                   | Wittig reaction. Mannich reaction, *Michael reaction. * Use of acetals as protecting group.   |
| 26thFebruary-02nd                        | Oxidation of aldehydes, Baeyer-Villiger oxidation of ketones, Cannizzaro reaction. MPV,   |
| March                                    | Clemmensen, Wolff-Kishner, LiAlH4 and NaBH4 reductions. * Halogenation of enolizable  |
| 03 <sup>th</sup> March                   | ketones. *An introduction to α, β-unsaturated aldehydes and ketones.  SUNDAY  |
| Week 8                                   | Revision, Assignment and Test   |
| 04 <sup>th</sup> March-09 <sup>th</sup>  | Revision, Assignment and Test   |
| 10 <sup>th</sup> March                   | SUNDAY  |
| Week 9                                   | Amines  |
| 11 <sup>th</sup> March–16 <sup>th</sup>  | Structure and nomenclature of amines, physical properties. Stereochemistry of amines. Separation  |
| March                                    | of a mixture of primary, secondary and tertiary amines. Structural features affecting basicity of   |
| March                                    | amines. *Amine salts as phase-transfer catalysts.   |
| 17 <sup>th</sup> March                   | SUNDAY  |
| Week 10                                  | Preparation of alkyl and aryl amines (reduction of nitro compounds, nitriles, reductive amination   |
| 18 <sup>th</sup> March– 22 <sup>th</sup> | of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide reaction.  |
| March                                    | Electrophilic aromatic substitution in aryl amines, reactions of amines with nitrous acid.  |
| 24 March                                 | SUNDAY  |
| Week 11                                  | 23-31 March Holi Break  |
| Week 12                                  | Diazonium Salts   |
| 01st April - 06th                        | Mechanism of diazotisation, structure of benzene diazonium chloride, Replacement of diazo group   |
| April                                    | by H, OH, F, Cl, Br, I, NO2 and CN groups, reduction of diazonium salts to hyrazines, coupling  |
|  | reaction and its synthetic application. * Preparation and reactions of cyanides, and isocyanides,   |
| 0  | urea and diazomethane.  |
| 07 <sup>th</sup> April                   | SUNDAY  |
| Week 13                                  | Nitro Compounds   |
| 8 <sup>th</sup> April – 13 <sup>th</sup> | Preparation of nitro alkanes and nitro arenes and their chemical reactions. Mechanism of  |
| April                                    | electrophilic substitution reactions in nitro arenes and their reductions in acidic, neutral and alkaline medium. *Picric acid. Halonitroarenes: reactivity, Preparation of nitro alkanes and nitro |
|  | arenes  |
| 14 <sup>th</sup> April                   | SUNDAY  |
| Week 14                                  | Chemical reactions. Mechanism of electrophilic substitution reactions in nitro arenes and their   |
| 15 <sup>th</sup> April –                 | reductions in acidic, neutral and alkaline medium. *Picric acid. Halonitroarenes: reactivity  |
| 20 <sup>th</sup> April                   | ,   |
| 21 <sup>st</sup> April                   | SUNDAY  |
|  |   |
| Wook 15                                  | Priof Section A. P. C. D. Oversieg/ Develop   |
| Week 15                                  | Brief Section A, B, C, D Queries/ Doubts  |
| 22 <sup>nd</sup> April – 27              |   |
| April                                    | CULINITY A NA   |
| 28 <sup>th</sup> April                   | SUNDAY  |
| *** 1.46                                 |   |
| Week 16                                  | Brief Section A, B, C, D Queries/ Doubts  |
| 29 <sup>th</sup> April-30 <sup>th</sup>  |   |
| April                                    |   |

| Name of the Assistant/Associate Professor: Anita Amani<br>Class and Section: B.Sc. Chemistry Hons. (VI semester) |  |  |  |
|--|--|--|--|
| Subject: Chemistry<br>Paper: Inorganic Chemi   | istry Paper-2  |  |  |
| 1 <sup>st</sup> January – 15 <sup>t</sup> h<br>January   | Practical Slot   |  |  |
| Week 1   | Overview of syllabus, General Introduction   |  |  |
| 16 January - 20<br>January   | Primary and secondary pollutants, sources, pollution effects   |  |  |
| 21st January   | SUNDAY   |  |  |
| Week 2<br>22 <sup>rd</sup> January – 27 <sup>th</sup><br>January   | AIR POLLUTION control of the following. Gaseous hydrocarbons, carbon monoxide, carbon dioxide, hydrogen sulfide.   |  |  |
| 28 <sup>th</sup> January   | SUNDAY   |  |  |
| Week 3  29 <sup>th</sup> January - 03 <sup>th</sup> February   | AIR PURIFICATION  Air purification by microorganisms, Acid rain, Mechanism of photochemical smog formation.  oxides of sulfur and nitrogen and ozone.  |  |  |
| 04 <sup>th</sup> February  | SUNDAY   |  |  |
| Week 4  05 <sup>th</sup> February 10 <sup>th</sup> February  | INDUSTRIAL WASTE & TREATMENT PROCESS Introduction, characteristics of industrial wastes, types of industrial wastes, principles of industrial waste treatment and disposal of industrial wastes. |  |  |
| 11 <sup>th</sup> February  | SUNDAY   |  |  |
| Week 5<br>12 <sup>th</sup> February – 17t <sup>h</sup><br>February   | NUCLEAR & RADIOCHEMISTRY  Composition of Nuclei, structure of nucleus, forces operative within nucleus, nuclear stability and mass energy equivalence (binding energy).                          |  |  |

| 18th February  | SUNDAY  |  |  |  |
|--|---|--|--|--|
| Week 6   | TYPES OF NUCLEAR REACTIONS  |  |  |  |
| 19 <sup>th</sup> February – 24 <sup>th</sup><br>February | The compound nucleus theories, thermonuclear reactions including  |  |  |  |
| 25 <sup>th</sup> February                                | SUNDAY  |  |  |  |
| Week 7   | RADIATION DETECTION & MEASUREMENT Gascous ion collection Methods (G.M., ionization and proportional counters) |  |  |  |
| 26thFebruary-02nd<br>March                               | scintillation counter, semiconductors detectors   |  |  |  |
| 03 <sup>th</sup> March                                   | SUNDAY  |  |  |  |
| Week 8   | TRACERS IN CHEMISTRY  |  |  |  |
| 04 <sup>th</sup> March-09 <sup>th</sup>                  | Activation analysis, isotope dilution analysis and radiometric titrations.                                    |  |  |  |
| 10 <sup>th</sup> March                                   | SUNDAY  |  |  |  |
| Week 9   | WATER POLLUTION   |  |  |  |
| 11 <sup>th</sup> March–16 <sup>th</sup> March            | Types of water pollution, sources of water pollution, approaches to prevent and control water pollution       |  |  |  |
| 17 <sup>th</sup> March                                   | SUNDAY  |  |  |  |
| Week 10  | <u>CRYSTAL STRUCTURE</u>  |  |  |  |
| 18 <sup>th</sup> March– 22 <sup>th</sup> March           | Structures of binary compounds such as zinc blende, wurtzite, NiAs, CsCl, CaF2, rutile,                       |  |  |  |
| 24 March   | SUNDAY Week 11 23-31 March Holi Break   |  |  |  |
| Week 12  | Revision  |  |  |  |
| 01st April – 06th April                                  |   |  |  |  |
| 07 <sup>th</sup> April                                   | SUNDAY  |  |  |  |
| Week 13  | Revision  |  |  |  |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April           |   |  |  |  |
| 14 <sup>th</sup> April                                   | SUNDAY  |  |  |  |
| Week 14  | Revision  |  |  |  |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April          |   |  |  |  |
| 21stApril  | SUNDAY  |  |  |  |

| Week 15                                       | Revision |
|---|----------|
| 22 <sup>nd</sup> April – 27 April             |          |
| Week 16                                       | Revision |
| 29 <sup>th</sup> April-30 <sup>th</sup> April |          |

| Summary of Lesson Plan of College Faculty Name of College: Pt. Neki Ram Sharma Government College, Rohtak Semester: Even Name of Asstt./Ass. ProF. – Abhishek Dahiya Class: B.Sc Chemistry Hons. 6 <sup>th</sup> SEM  PaperLII Inorganic Chemistry (Theory)  1st January – 15 <sup>th</sup> Practical Slot January Overview of syllabus, General Introduction |  |  |
|---|--|--|
| 16 January - 20<br>January  |  |  |
| 21st January  | SUNDAY   |  |
| Week 2<br>22 <sup>rd</sup> January – 27 <sup>th</sup><br>January  | Criteria for choice of organic reagents, use of following reagents in inorganic analysis   |  |
| 28 <sup>th</sup> January  | SUNDAY   |  |
| Week 3  29 <sup>th</sup> January - 03 <sup>th</sup> February  | DMG, cupferron, 8-hydroxyquinoline, Nitroso □- naphthol  |  |
| 04 <sup>th</sup> February   | SUNDAY   |  |
| Week 4<br>05 <sup>th</sup> February 10 <sup>th</sup><br>February  | EDTA, Acetylacetone, dithiozone, dithiozone, dithiozone, disadvantages and disadvantages of organic reagents in inorganic analysis |  |
| 11 <sup>th</sup> February   | SUNDAY   |  |
| Week 5<br>12 <sup>th</sup> February – 17t <sup>h</sup><br>February  | Analytical Chemistry: Sources of errors in chemical analysis, classification of errors   |  |
| 18th February   | SUNDAY   |  |
| Week 6<br>19 <sup>th</sup> February – 24 <sup>th</sup><br>February  | precision, accuracy, statistical evaluation and interpretation of results in analytical chemistry (with numericals).               |  |
| 25 <sup>th</sup> February   | SUNDAY   |  |
| Week 7<br>26thFebruary–02nd<br>March  | Inorganic Polymers: Definition, classification, polymers based on hetroatomic structure  |  |

| 03 <sup>th</sup> March                                       | SUNDAY  |  |
|--|---|--|
| Week 8<br>04 <sup>th</sup> March–09 <sup>th</sup>            | PON polymer, polythiazyl, synthetic inorganic fibres Co-ordination polymers.  |  |
| 10 <sup>th</sup> March                                       | SUNDAY  |  |
| Week 9  11 <sup>th</sup> March–16 <sup>th</sup> March        | Solvent Extraction: Basic principles of solvent extraction, classification and mechanism of extraction                      |  |
| 17 <sup>th</sup> March                                       | SUNDAY  |  |
| Week 10<br>18 <sup>th</sup> March– 22 <sup>th</sup><br>March | extraction equilibria, techniques of extraction and applications in analytical chemistry.                                   |  |
| 24 March   | SUNDAY Week 11 23-31 March Holi Break   |  |
| Week 12 01st April – 06th April                              | Ion - Exchange:<br>Characteristics of ion-exchangers, mechanism of ion-exchange, ion-exchange equilibra                     |  |
| 07 <sup>th</sup> April                                       | SUNDAY  |  |
| Week 13<br>8 <sup>th</sup> April – 13 <sup>th</sup> April    | plate theory for ion-exchange, techniques of ion-exchange and applications of ion exchange for separations., (Assignment 2) |  |
| 14 <sup>th</sup> April                                       | SUNDAY  |  |
| Week 14  15 <sup>th</sup> April – 20 <sup>th</sup> April     | Chromatography: Classification of chromatographic methods, chromatographic terminology - Rf value, partition co-efficient   |  |
| 21stApril  | SUNDAY  |  |
| Week 15 22 <sup>nd</sup> April – 27 April                    | dyanmics of chromatography, basic principles of adsorption and partition chromatography, applications                       |  |
| 28 <sup>th</sup> April                                       | SUNDAY  |  |
| Week 16  | Brief Section A, B, C, D Queries/ Doubts  |  |
| 29 <sup>th</sup> April-30 <sup>th</sup> April                |   |  |

Paper- LV (Theory) Physical Chemistry -II

| Summary of Lesson Plan of College Facult | Summary | of Lesson | Plan of | College | Faculty |
|--|---------|-----------|---------|---------|---------|
|--|---------|-----------|---------|---------|---------|

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2023-24 **Semester: Even** 

Name of Asstt./Ass. ProF. – Abhishek Dahiya Class: B.Sc Chemistry Hons. 6<sup>th</sup> SEM

| Name of Subject: Unemis | of Subject: Chemistr | rv |
|-------------------------|----------------------|----|
|-------------------------|----------------------|----|

| Name of Subject: Chem.   | isti y  |
|--|---|
| 1 <sup>st</sup> January – 15 <sup>t</sup> h<br>January             | Practical Slot  |
| Week 1<br>16 January - 20<br>January                               | Overview of syllabus, General Introduction  |
| 21st January   | SUNDAY  |
| Week 2   | Planck's law, heat capacity of solids   |
| 22 <sup>rd</sup> January – 27 <sup>th</sup><br>January             |   |
| 28 <sup>th</sup> January   | SUNDAY  |
| Week 3  29 <sup>th</sup> January - 03 <sup>th</sup> February       | Bohr's model of hydrogen atom (derivation excluded) and its defects.)                                       |
| 04 <sup>th</sup> February  | SUNDAY  |
| Week 4  05 <sup>th</sup> February 10 <sup>th</sup> February        | Compton effect, molecular orbital theory, basic idea. Compton effect, molecular orbital theory, basic idea. |
| 11 <sup>th</sup> February  | SUNDAY  |
| Week 5<br>12 <sup>th</sup> February – 17t <sup>h</sup><br>February | Construction of molecular orbital by linear combination of atomic orbital ,-H2 ion.                         |
| 18th February  | SUNDAY  |
| Week 6<br>19 <sup>th</sup> February – 24 <sup>th</sup><br>February | Calculation of energy levels from wave function, physical picture of bonding and antibonding wave function. |
| 25 <sup>th</sup> February  | SUNDAY  |
| Week 7 26thFebruary-02nd   | Concept of pie, pie* orbitals and their characteristics. Hybrid orbital (Sp. Sp2 and Sp3.)                  |

| March  |  |  |
|--|--|--|
| 03 <sup>th</sup> March                                       | SUNDAY   |  |
| Week 8<br>04 <sup>th</sup> March–09 <sup>th</sup>            | Calculation of co-efficients of atomic orbitals used in these hybrid orbitals. Introduction of valence bond model of H2,comparison of molecular orbital. and valence bond. model |  |
| 10 <sup>th</sup> March                                       | SUNDAY   |  |
| Week 9   | Homogeneous and Heterogeneous catalysis, Enzyme catalysis. Theory of catalysis -   |  |
| 11 <sup>th</sup> March–16 <sup>th</sup><br>March             | Intermediate compound formation theory, adsorption theory  |  |
| 17 <sup>th</sup> March                                       | SUNDAY   |  |
| Week 10<br>18 <sup>th</sup> March– 22 <sup>th</sup><br>March | General characteristics of catalysis, positive catalysis, negative catalysis, autocatalysis, shape selective catalysis.  |  |
| 24 March   | SUNDAY Week 11 23-31 March Holi Break  |  |
| Week 12 01st April – 06th April                              | Classification of chromatographic methods, principle of differential migration, nature of differential migration   |  |
| 07 <sup>th</sup> April                                       | SUNDAY   |  |
| Week 13  | Adsorption phenomenon, nature of adsorpent, solvent system. Rf- values, (Assignment 2)   |  |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April               |  |  |
| 14 <sup>th</sup> April                                       | SUNDAY   |  |
| Week 14  | application basic principle of partition, paper, column, thin layer liquid-liquid partition and high performance.  |  |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April              |  |  |
| 21stApril  | SUNDAY   |  |
| Week 15  | Liquid chromatography, paper & column, thin layer  |  |
| 22 <sup>nd</sup> April – 27 April                            | liquid-liquid partition and high performance liquid chromatography  Submission of assignments and Queries will be taken.   |  |
| 28 <sup>th</sup> April                                       | SUNDAY   |  |
| Week 16  | Brief Section A, B, C, D Queries/ Doubts   |  |
| 29th April-30th April  |  |  |

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2023-24

Semester: Even

Name of Asstt./Ass. ProF. -Dr. Rinki

Class: B.Sc Chemistry Hons. 6th SEM Paper- LVII (Theory) Organic Chemistry -II

| 1 <sup>st</sup> January – 15 <sup>t</sup> h     | Practical Slot   |
|---|--|
| · ·   | Fractical Stot   |
| January   |  |
| Week 1  | Overview of syllabus, General Reactions Introduction   |
| 16 January - 20                                 | Overview of synabus, General Reactions introduction  |
| January - 20                                    |  |
| 21st January                                    | SUNDAY   |
| Week 2  | Introduction, essential oils, classification of terpenes, Isoprene rule, isolation                         |
| 22 <sup>rd</sup> January – 27 <sup>th</sup>     | introduction, essential ons, classification of terpenes, isoprene rule, isolation                          |
| January   |  |
| 28 <sup>th</sup> January                        | SUNDAY   |
| Week 3  | General structure elucidation, Structure elucidation and synthesis of citral and geraniol. ( <b>Test</b> ) |
| 29 <sup>th</sup> January - 03 <sup>th</sup>     | General structure crucidation, Structure crucidation and synthesis of crual and geranior. (16st)           |
| February - 03                                   |  |
| 04 <sup>th</sup> February                       | SUNDAY   |
| Week 4  | Introduction, classification, extraction, Physiological action in alkaloids, general characteristics,      |
| 05 <sup>th</sup> February 10 <sup>th</sup>      | general methods of determining structures  |
| February  | Sentent intented of determining burdenies  |
| 11 <sup>th</sup> February                       | SUNDAY   |
| Week 5  | Hofmann's exhaustive methylation, isolation, structure elucidation and synthesis of nicotine               |
| 12 <sup>th</sup> February – 17t <sup>h</sup>    | Tromain 5 exhaustive methylation, isolation, structure elaboration and synthesis of meetine                |
| February  |  |
| 18th February                                   | SUNDAY   |
| Week 6  | Structure elucidation and synthesis of cocaine, coniine and Piperine (Assignment 1)                        |
| 19 <sup>th</sup> February – 24 <sup>th</sup>    |  |
| February  |  |
| 25 <sup>th</sup> February                       | SUNDAY   |
| Week 7  | Classification, Natural pesticides: Nicotinides, Pyrethroids, Rotenoide                                    |
| 26thFebruary-02nd                               |  |
| March   |  |
| 03 <sup>th</sup> March                          | SUNDAY   |
| Week 8  | Sabodilia, Ryania, Synthetic pesticides: Nitrophenols, Halogens derivatives of aromatic                    |
| 04th March-09th                                 | hydrocarbons and alicyclic hydrocarbon   |
| 10 <sup>th</sup> March                          | SUNDAY   |
| Week 9  | Organo phosphorus pesticides. Preparation, reactions and uses of DDT, BHC, Malathion and                   |
| 11 <sup>th</sup> March-16 <sup>th</sup>         | Parathion (Test)   |
| March   |  |
| 17 <sup>th</sup> March                          | SUNDAY   |
| Week 10   | Introduction, classification, pro vitamins, occurrence, structure  |
| 18 <sup>th</sup> March– 22 <sup>th</sup>        |  |
| March   |  |
| 24 March  | SUNDAY Week 11 23-31 March Holi Break  |
| Week 12   | Deficiency diseases of vitamins A, B complex (B 1 ,B2 , B6 and B1 2 ), C, D, E, H and K                    |
| 01 <sup>st</sup> April – 06 <sup>th</sup> April | 2 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -  |
| 07 <sup>th</sup> April                          | SUNDAY   |
| ~, 1 <b>1</b> P111                              |  |

| Week 13  | Introduction, functions, difference between hormones and vitamins, classification and study of |
|--|--|
| 8 <sup>th</sup> April – 13 <sup>th</sup> April           | Thyroxine (Assignment 2)   |
| 14 <sup>th</sup> April                                   | SUNDAY   |
| Week 14  | Adrenalin, Insulin, Testosterone, Progesterone (structure, secreting gland and functions)      |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April          |  |
| 21stApril  | SUNDAY   |
|  |  |
| Week 15  | Estrogens, Cortison hormone (structure, secreting gland and functions)                         |
| 22 <sup>nd</sup> April – 27 April                        | Submission of assignments and Queries will be taken.   |
| 28 <sup>th</sup> April                                   | SUNDAY   |
| Week 16<br>29 <sup>th</sup> April-30 <sup>th</sup> April | Brief Section A, B, C, D Queries/ Doubts   |

| Name of College: Pt. Nel<br>Name of Asstt./Ass. Prof                                   | ki Ram Sharma Government College, Rohtak Academic Session 2023-24 Semester: Even  |  |
|--|---|--|
| Class: B.Sc Chemistry Hons. 6 <sup>th</sup> SEM  Name of Subject: Physical Chemistry I |   |  |
|  |   |  |
| Week 1<br>16 January - 20<br>January<br>21st January                                   | Vibrational Spectroscopy: Infrared spectrum: energy levels of simple harmonic oscillator. Selection rule. Pure vibration spectrum. Rotational - vibration spectrum. Calculation of energy of levels and selection rule SUNDAY |  |
| Week 2<br>22 <sup>rd</sup> January – 27 <sup>th</sup><br>January                       | Vibrational Spectroscopy: Intensity determination of force constant and qualitative relation of force constant and bond energies. Effect of anharmonic motions and isotope on the spectrum.                                   |  |
| 28 <sup>th</sup> January   | SUNDAY  |  |
| Week 3 29 <sup>th</sup> January - 03 <sup>th</sup> February                            | Vibrational Spectroscopy: Idea of vibrational frequencies of different functional group. Rotational - vibration spectrum. Calculation of energy of levels and selection rule.   |  |
| 04 <sup>th</sup> February  | SUNDAY  |  |
| Week 4<br>05 <sup>th</sup> February 10 <sup>th</sup><br>February                       | Quantum Mechanics: Dual nature of matter and light. Photoelectric effect, De-Broglie equation. Heirsenberg's uncertainty principle, Schrodinger wave equation and its significance.   |  |
| 11 <sup>th</sup> February  | SUNDAY  |  |
| Week 5<br>12 <sup>th</sup> February – 17t <sup>h</sup><br>February                     | Quantum Mechanics: Physical interpretation of the wave function. Postulates of quantum mechanics.   |  |
| 18th February  | SUNDAY  |  |
| Week 6<br>19 <sup>th</sup> February – 24 <sup>th</sup><br>February                     | Quantum Mechanics: Particle in one dimensional box, Particle in three dimensional box and concept of degenracy  |  |
| 25 <sup>th</sup> February  | SUNDAY  |  |
| Week 7<br>26t <sup>h</sup> February–02 <sup>nd</sup><br>March                          | Revision, Assignment and Test   |  |
| 03 <sup>th</sup> March   | SUNDAY  |  |
| Week 8<br>04 <sup>th</sup> March–09 <sup>th</sup>                                      | Raman Spectroscopy: Quantum theory of Raman effect. Classical theory of Raman effect. Pure rotational Raman spectra, Raman activity of vibration.   |  |
| 10 <sup>th</sup> March   | SUNDAY  |  |

| Week 9  | Raman Spectroscopy:   |
|---|---|
| 11 <sup>th</sup> March–16 <sup>th</sup>         | Vibration Raman spectra. Rotation - vibration Raman spectrum. Polarisation of light and Raman   |
| March   | effect. Experimental technique. Application of Raman effect.                                    |
| 17 <sup>th</sup> March                          | SUNDAY  |
|   | SUNDAT  |
| Week 10   |   |
| 18 <sup>th</sup> March– 22 <sup>th</sup>        | Elementary idea of NMR, Coupling Constant, Chemical Shift                                       |
| March   |   |
| 24 March  | SUNDAY  |
| Week 11   | 23-31 March Holi Break  |
| Week 12   | Electronic Spectra:   |
| 01st April - 06th                               | Concepts of potential energy curves for bonding and antibonding molecular orbitals. Qualitative |
| April   | description of selection rule. Franck-condon principle.   |
| 07 <sup>th</sup> April                          | SUNDAY  |
| Week 13   | Electronic Spectra:   |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April  | Qualitative description of , PIE and DELTA orbitals and their energy level and their respective |
| 8 April - 13 April                              | transition.   |
| 14 <sup>th</sup> April                          | SUNDAY  |
| Week 14   | SUNDAT  |
|   |   |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April | Elementary ideas of Electron Spin Resonance Spectroscopy. Application ESR spectroscopy          |
| 21stApril                                       | SUNDAY  |
|   |   |
| Week 15   | Revision, Assignment and Test   |
| 22 <sup>nd</sup> April – 27 April               |   |
|   |   |
| 28 <sup>th</sup> April                          | SUNDAY  |
| _   |   |
| XX/1-1/   | Davisian Assignment and Test  |
| Week 16   | Revision, Assignment and Test   |
| 29 <sup>th</sup> April-30 <sup>th</sup> April   |   |

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2023-2024

**Semester: Even** 

Name of Asstt./Ass. ProF. – Anil, Lokesh Kumar Class: B.Sc. Non Medical Sec. A,B,C 2<sup>nd</sup> Sem.

Name of Subject: Inorganic Chemistry

| 1 <sup>st</sup> January – 15 <sup>t</sup> h Practical Slot<br>January   |                   |
|---|-------------------|
|   |                   |
| Week 1  Hydrogen Bonding & Vander Waals Forces Hydrogen Bonding – Definition, hydrogen bonding on properties of substances                          | Types, effects of |
| 21st January SUNDAY   |                   |
| Week 2 application Brief discussion of various types of Vander Waals Forces  22 <sup>rd</sup> January – 27 <sup>th</sup> January                    |                   |
| 28th January SUNDAY   |                   |
| Week 3  29 <sup>th</sup> January - 03 <sup>th</sup> February  Metallic Bond and Semiconductors  Metallic Bond- Brief introduction to metallic bond, |                   |
| 04th February SUNDAY  |                   |
| Week 4 band theory of metallic bond Semiconductors- Introduction, types and application of the February February 10 <sup>th</sup> February          | ns.               |
| 11 <sup>th</sup> February SUNDAY  |                   |
| Week 5 . s-Block Elements Comparative study of the elements including , diagonal relation 12 <sup>th</sup> February – 17t <sup>h</sup> February     | nships,           |
| 18th February SUNDAY  |                   |
| Week 6  19 <sup>th</sup> February – 24 <sup>th</sup> February   | r function in     |
| 25 <sup>th</sup> February SUNDAY  |                   |

| Week 7<br>26thFebruary-02nd<br>March                  | Chemistry of Noble Gases Chemical properties of the noble gases with emphasis on their low chemical reactivity, chemistry of xenon, structure and bonding of fluorides, oxides & oxyfluorides of xenon   |
|---|--|
| 03 <sup>th</sup> March                                | SUNDAY   |
| Week 8<br>04 <sup>th</sup> March–09 <sup>th</sup>     | Boron family (13th gp):- Diborane – properties and structure (as an example of electron – deficient compound and multicentre bonding), Borazene – chemical properties and structure Trihalides of Boron – Trends in fewis acid character structure of aluminium (III) chloride |
| 10 <sup>th</sup> March                                | SUNDAY   |
| Week 9  11 <sup>th</sup> March–16 <sup>th</sup> March | Carbon Family (14th group) Catenation, p $\pi$ – d $\pi$ bonding (an idea), carbides, fluorocarbons, silicates structural aspects), silicons – general methods of preparations, properties and uses  |
| 17 <sup>th</sup> March                                | SUNDAY   |
| Week 10   | Nitrogen Family (15th group) Oxides – structures of oxides of N,P.   |
| 18 <sup>th</sup> March– 22 <sup>th</sup><br>March     | test   |
| 24 March  | SUNDAY Week 11 23-31 March Holi Break  |
| Week 12<br>01st April – 06th April                    | oxyacids – structure and relative acid strengths of oxyacids of Nitrogen and phosphorus. Structure of white, yellow and red phosphorus   |
| 07 <sup>th</sup> April                                | SUNDAY   |
| Week 13   | Oxygen Family (16th group) Oxyacids of sulphur – structures  |
| $8^{th}\ April-13^{th}\ April$                        |  |
| 14 <sup>th</sup> April                                | SUNDAY   |
| Week 14   | acidic strength H2O2 –structure, properties and uses.  |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April       |  |
| 21stApril   | SUNDAY   |
| Week 15  22 <sup>nd</sup> April – 27 April            | Halogen Family (17th group) Basic properties of halogen, interhalogens types properties  |
| 28 <sup>th</sup> April                                | SUNDAY   |
| Week 16 29 <sup>th</sup> April-30 <sup>th</sup> April | oxyacids of chlorine – structure and comparison of acid strength   |

Name of College: Pt. Neki Ram Sharma Government College, Rohtak

Academic Session 2023-24 Semester: Even Name of Asstt./Ass. ProF. – Praveen & Ruman Rani Class: B.Sc. Medical & N.M Sec A,B,C 2<sup>nd</sup> Sem.

| Week-1,2 (1Jan-           | Practical schedule  |
|---------------------------|---|
| 15Jan)                    |   |
| Week-3 (16Jan-            | Chemical kinetics and its scope, Rate of reaction, factors influencing the rate of reaction.                                |
| 20Jan)                    | Concentration, temperature, pressure, solvent, light, catalyst, concentration dependence of                                 |
|                           | rates. Mathematical characteristics of simple chemical reactions  |
| Week-4 (22Jan-            | Molecularity and order of reaction. Zero order ,lst order ,second order, third order reactions                              |
| 27Jan)                    | and their mathematical derivations for their rate constants.  |
| 28Jan                     | SUNDAY  |
| Week 5<br>(29Jan-3Feb)    | Half life period, average life period, determination of order reaction.   |
| 4 Feb                     | SUNDAY  |
| Week 2<br>(5Feb-10Feb)    | Differential method, method of integration. Method of half-life period and isolation method. Pseudo uni molecular reactions |
| Week 2<br>(5Feb-10Feb)    | Effect of temperature of rate of raction-Arrhenius equation   |
| 11 Feb                    | SUNDAY  |
| Week 3<br>(12 Feb-17Feb)  | Theories of reaction rates  |
| 18th Feb                  | SUNDAY  |
| Week 4                    | Electrochemistry-I Electrical transport conduction in metal and in electrolyte solutions, specific                          |
| (19 Feb-24Feb)            | conductance and equivalent conductance. Measurement of equivalent conductance.  |
| Week 5<br>(26 Feb-2March) | Variation of equivalent conductance and specific conductance with dilution, migration of ions                               |
| 3 March                   | SUNDAY  |

| Week 2            | Kohlrausch's law, Arhenious theory of electrolyte dissolution and its limitations.                     |
|-------------------|--|
| 4 March-9 March   |  |
| 10March           | SUNDAY   |
| Week 3            | Weak and strong electrolytes. Ostwald's dilution law and its uses and limitation                       |
| (11March-16March) |  |
| 17 March          | SUNDAY   |
| Week 4            | Electrochemistry-II Debye-Huckel onsager equation for strong electrolytes (elementary treatment        |
| 18March- 22March  | only),transport number   |
|                   |  |
|                   |  |
| 23March-31March   | Holi -Break (Holiday)  |
| Week 1            | Hittorf and moving boundary method. Application of conductivity measurements                           |
| 1April-6 April    |  |
|                   |  |
| 7 April           | SUNDAY   |
| Week 2            | Determination of solubility product of sparingly soluble salts. Determination of degree of dissolution |
| 8April-13 April   | ,Ka for weak acids   |
| 14 April          | SUNDAY   |
|                   |  |
| Week 3            | Buffer solutions, buffer actions   |
| 15April-20 April  |  |
| 21 April          | SUNDAY   |
| Week 4            | Henderson.Hazel, buffer mechanism of buffer action   |
| 22April-27 April  |  |
| 28 April          | SUNDAY   |
| Week 4            | Test   |
| 29April-30 April  |  |

Summary of Lesson Plan of College Faculty Name of College: Pt. Neki Ram Sharma Government College, Rohtak

**Academic Session 2023-24** 

Semester: Even

Name of Asstt./Ass. Prof. – Nidhi Mann & Jyoti Dalal Class: B.Sc Pass 2<sup>nd</sup> SEM [Med & Non Med]

**Subject- organic Chemistry** 

| 1 <sup>st</sup> January – 15 <sup>t</sup> h   | Practical Slot  |
|---|---|
| January   |   |
|   | Nomenclature of alkenes, , mechanisms of dehydration of alcohols and  |
| Week 1  | dehydrohalogenation of alkyl halides,. The Saytzeff rule, Hofmann elimination, physical p   |
| 16 January - 20 January   | roperties and relative stabilities of alkenes   |
| 21st January  | SUNDAY  |
| Week 2  | s. Chemical reactions of alkenes mechanisms involved in hydrogenation, electrophilic and  |
| 22 <sup>rd</sup> January – 27 <sup>th</sup>   | free radical additions,   |
| January   |   |
| 28 <sup>th</sup> January  | SUNDAY  |
| Week 3  | Markownikoff's rule, hydroboration-oxidation, oxymercurationreduction, ozonolysis,  |
| 29 <sup>th</sup> January - 03 <sup>th</sup>   | hydration, hydroxylation and oxidation with KMnO4,  |
| February  | CANADA A V  |
| 04 <sup>th</sup> February   | SUNDAY  |
| Week 4<br>05 <sup>th</sup> February 10 <sup>th</sup>  | Chemical reactions of alkenes mechanisms involved in hydrogenation, electrophilic and   |
| February 10 <sup>th</sup>   | free radical additions, Markownikoff's rule, hydroboration—oxidation,   |
|   | oxymercurationreduction, ozonolysis, hydration, hydroxylation and oxidation with KMnO4  |
| 11 <sup>th</sup> February   | SUNDAY  |
| Week 5  |   |
| 12 <sup>th</sup> February – 17t <sup>h</sup>  | Nomenclature of benzene derivatives:. Aromatic nucleus and side chain. Aromaticity: the   |
| February  | Huckel rule, aromatic ions, annulenes up to 10 carbon atoms, aromatic, anti-aromatic and  |
|   | non-aromatic compounds.   |
|   | P   |
|   |   |
| 18th February   | SUNDAY  |
| Week 6  |   |
| Week 6<br>19 <sup>th</sup> February – 24 <sup>th</sup>  | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of  |
| Week 6  | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams.   |
| Week 6<br>19 <sup>th</sup> February – 24 <sup>th</sup>  | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of  |
| Week 6<br>19 <sup>th</sup> February – 24 <sup>th</sup><br>February  | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams.   |
| Week 6<br>19 <sup>th</sup> February – 24 <sup>th</sup>  | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February   | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February Week 7  | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes.   |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February Week 7 26t <sup>h</sup> February–02 <sup>nd</sup>   | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic &  |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February Week 7 26t <sup>h</sup> February–02 <sup>nd</sup>   | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes.   |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February  Week 7 26t <sup>h</sup> February–02 <sup>nd</sup> March  | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction  |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February  Week 7 26thFebruary–02nd  March  03 <sup>th</sup> March  | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction  SUNDAY  |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February  Week 7 26t <sup>h</sup> February–02 <sup>nd</sup> March  | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction  |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February  Week 7 26t <sup>h</sup> February–02 <sup>nd</sup> March  03 <sup>th</sup> March  Week 8  | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction  SUNDAY  |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February Week 7 26thFebruary–02 <sup>nd</sup> March  03 <sup>th</sup> March Week 8 04 <sup>th</sup> March–09 <sup>th</sup>   | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction  SUNDAY  Revision and Test of unit I and unit II   |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February Week 7 26t <sup>h</sup> February–02 <sup>nd</sup> March  03 <sup>th</sup> March Week 8 04 <sup>th</sup> March–09 <sup>th</sup> 10 <sup>th</sup> March                             | Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction  SUNDAY  Revision and Test of unit I and unit II  SUNDAY  Nomenclature, structure and bonding in alkynes. Methods of formation. Chemical reactions   |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February Week 7 26thFebruary–02 <sup>nd</sup> March  03 <sup>th</sup> March Week 8 04 <sup>th</sup> March–09 <sup>th</sup> 10 <sup>th</sup> March Week 9                                   | Aromatic electrophilic substitution general pattern of the mechanism, mechansim of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction  SUNDAY  Revision and Test of unit I and unit II  SUNDAY  Nomenclature, structure and bonding in alkynes. Methods of formation. Chemical reactions of alkynes, acidity of alkynes. Mechanism of electrophilic and nucleophilic addition  |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February Week 7 26thFebruary–02 <sup>nd</sup> March  03 <sup>th</sup> March Week 8 04 <sup>th</sup> March–09 <sup>th</sup> 10 <sup>th</sup> March Week 9                                   | Aromatic electrophilic substitution general pattern of the mechanism, mechansim of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction  SUNDAY  Revision and Test of unit I and unit II  SUNDAY  Nomenclature, structure and bonding in alkynes. Methods of formation. Chemical reactions   |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February Week 7 26thFebruary–02nd March  03 <sup>th</sup> March Week 8 04 <sup>th</sup> March–09 <sup>th</sup> 10 <sup>th</sup> March Week 9 11 <sup>th</sup> March–16 <sup>th</sup> March | Aromatic electrophilic substitution general pattern of the mechanism, mechansim of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction  SUNDAY  Revision and Test of unit I and unit II  SUNDAY  Nomenclature, structure and bonding in alkynes. Methods of formation. Chemical reactions of alkynes, acidity of alkynes. Mechanism of electrophilic and nucleophilic addition reactions, hydroboration- oxidation of alkynes |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February Week 7 26thFebruary–02nd March  03 <sup>th</sup> March Week 8 04 <sup>th</sup> March–09 <sup>th</sup> 10 <sup>th</sup> March Week 9 11 <sup>th</sup> March–16 <sup>th</sup> March | Aromatic electrophilic substitution general pattern of the mechanism, mechansim of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction  SUNDAY  Revision and Test of unit I and unit II  SUNDAY  Nomenclature, structure and bonding in alkynes. Methods of formation. Chemical reactions of alkynes, acidity of alkynes. Mechanism of electrophilic and nucleophilic addition  |
| Week 6 19 <sup>th</sup> February – 24 <sup>th</sup> February  25 <sup>th</sup> February Week 7 26thFebruary–02nd March  03 <sup>th</sup> March Week 8 04 <sup>th</sup> March–09 <sup>th</sup> 10 <sup>th</sup> March Week 9 11 <sup>th</sup> March–16 <sup>th</sup> March | Aromatic electrophilic substitution general pattern of the mechanism, mechansim of nitration, halogenation, sulphonation, and Friedel-Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.  SUNDAY  Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene,. Chemical reactions 1,2 and 1,4 additions (Electrophilic & free radical mechanism), Diels-Alder reaction  SUNDAY  Revision and Test of unit I and unit II  SUNDAY  Nomenclature, structure and bonding in alkynes. Methods of formation. Chemical reactions of alkynes, acidity of alkynes. Mechanism of electrophilic and nucleophilic addition reactions, hydroboration- oxidation of alkynes |

| 24 March   | SUNDAY Week 11 23-31 March Holi Break   |
|--|---|
| Week 12<br>01 <sup>st</sup> April – 06 <sup>th</sup> April | Mechanisms and stereochemistry of nucleophilic substitution reactions of alkyl halides, Sn2 and Sn1reactions with energy profile diagrams. Methods of formation and reactions of aryl halides |
| 07 <sup>th</sup> April                                     | SUNDAY  |
| Week 13<br>8 <sup>th</sup> April – 13 <sup>th</sup> April  | The additionelimination and the elimination-addition mechanisms of nucleophilic aromatic substitution reactions. Relative reactivities of alkyl halides vs allyl, vinyl and aryl halides.     |
| 14 <sup>th</sup> April                                     | SUNDAY  |
| Week 14<br>15 <sup>th</sup> April – 20 <sup>th</sup> April | Revision of unit III  |
| 21 <sup>st</sup> April                                     | SUNDAY  |
| Week 15<br>22 <sup>nd</sup> April – 27 April               | Revision of unit IV   |
| 28 <sup>th</sup> April                                     | SUNDAY  |
| Week 16<br>29 <sup>th</sup> April-30 <sup>th</sup> April   | Test of unit III and unit IV  |

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2023-24

**Semester: Even** 

Name of Asstt./Ass. Prof. - KIRAN, POONAM, NIDHI MANN

Class: B.Sc Pass 4<sup>th</sup> SEM [Non Med and Med.] Subject- Inorganic Chemistry

| 1 <sup>st</sup> January – 15 <sup>t</sup> h  | Practical Slot  |  |
|--|---|--|
| January – 13 h                               | Tractical Slot  |  |
| January                                      |   |  |
| Week 1                                       | Lanthanides   |  |
| 16 January - 20                              | Electronic structure, oxidation states and ionic radii and lanthanide contraction   |  |
| January                                      | Zavota vino su votavo, vinovinon sumos una rama una ruma una vinovinon  |  |
| 21st January                                 | SUNDAY  |  |
| Week 2                                       | lanthanide contraction, complex formation, lanthanide compounds.  |  |
| 22 <sup>rd</sup> January – 27 <sup>th</sup>  | r and |  |
| January                                      |   |  |
| 28th January                                 | SUNDAY  |  |
| Week 3                                       | occurrence and isolation  |  |
| 29 <sup>th</sup> January - 03 <sup>th</sup>  | (TEST)  |  |
| February                                     |   |  |
| 04 <sup>th</sup> February                    | SUNDAY  |  |
| Week 4                                       | Actinides   |  |
| 05 <sup>th</sup> February 10 <sup>th</sup>   | General features and chemistry of actinides   |  |
| February                                     |   |  |
| 11 <sup>th</sup> February                    | SUNDAY  |  |
| Week 5                                       | chemistry of separation of Np, Pu and   |  |
| 12 <sup>th</sup> February – 17t <sup>h</sup> | Am from U   |  |
| February                                     |   |  |
| 18th February                                | SUNDAY  |  |
| Week 6                                       | Comparison of properties of Lanthanides and Actinides and with  |  |
| 19 <sup>th</sup> February – 24 <sup>th</sup> | transition elements   |  |
| February                                     |   |  |
| 25 <sup>th</sup> February                    | SUNDAY  |  |
| Week 7                                       | . (ASSIGNMENT-1), TEST  |  |
| 26thFebruary-02nd                            | Submission of assignments and Queries will be taken.  |  |
| March  |   |  |
| 03 <sup>th</sup> March                       | SUNDAY  |  |
| Week 8                                       | Theory of Quali tative and Quanti tative Inorganic Analysis-I   |  |
| 04 <sup>th</sup> March–09 <sup>th</sup>      | Chemistry of analysis of various acidic radicals  |  |
| 4 0th 7 7                                    |   |  |
| 10 <sup>th</sup> March                       | SUNDAY  |  |
| Week 9                                       | Chemistry of identification of acid   |  |
| 11 <sup>th</sup> March–16 <sup>th</sup>      | radicals in typical combinations,   |  |
| March  | CUINIDAN7   |  |
| 17 <sup>th</sup> March                       | SUNDAY  Character of interference of a cite of including to the first   |  |
| Week 10                                      | Chemistry of interference of acid radicals including  |  |
| 18 <sup>th</sup> March— 22 <sup>th</sup>     | their removal in the analys is of basic radicals. (TEST)  |  |
| March  | CUNIDAY WI. 11 22 21 M 1 II P. 1  |  |
| 24 March                                     | SUNDAY Week 11 23-31 March Holl Break   |  |
| 24 March                                     | SUNDAY Week 11 23-31 March Holi Break   |  |

| Week 12  | Theory of Quali tative and Quanti tative Inorganic Analysis-II |
|--|--|
| 01 <sup>st</sup> April – 06 <sup>th</sup> April          | Chemistry of analysis of various groups of basic radicals,     |
| 07 <sup>th</sup> April                                   | SUNDAY   |
| Week 13  | Theory of precipitation, co- precipitation,                    |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April           |  |
| 14 <sup>th</sup> April                                   | SUNDAY   |
| Week 14  | Post- precipitation  |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April          |  |
| 21 <sup>st</sup> April                                   | SUNDAY   |
|  | purification of precipitates                                   |
| Week 15  | . (ASSIGNMENT-2)   |
| 22 <sup>nd</sup> April – 27 April                        | Submission of assignments and Queries will be taken.           |
| 28 <sup>th</sup> April                                   | SUNDAY   |
| Week 16<br>29 <sup>th</sup> April-30 <sup>th</sup> April | Brief Section A, B, C, D Queries/ Doubts                       |

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2023-24

**Semester: Even** 

Name of Asstt./Ass. Prof. – [A] KIRAN [B] ANIL[C] RINKU

Class: B.Sc Pass 4<sup>th</sup> SEM [Non Med and Med.] Subject-Physical Chemistry

| 1 <sup>st</sup> January – 15 <sup>t</sup> h  | Practical Slot  |
|--|---|
| January                                      |   |
| -  | Thermodynamics-III  |
| Week 1                                       | Second law of thermodynamics, need for the law, different statements of the law,  |
| 16 January - 20                              | Carnot's cycles and its efficiency, Carnot's theorm,  |
| January                                      |   |
| 21st January                                 | SUNDAY  |
| Week 2                                       | Thermodynamics scale of temperature. Concept of entropy – entropy as a state function, entropy as   |
| 22 <sup>rd</sup> January – 27 <sup>th</sup>  | a function of V & T, entropy as a function of P & T.  |
| January                                      |   |
| 28 <sup>th</sup> January                     | SUNDAY  |
| Week 3                                       | entropy change in physical change,  |
| 29 <sup>th</sup> January - 03 <sup>th</sup>  | entropy as a criteria of spontaneity and equilibrium. Entropy change in ideal gases   |
| February                                     | and mixing of gases (TEST)  |
| 04th February                                | SUNDAY  |
| Week 4                                       | Thermodynamics-IV   |
| 05 <sup>th</sup> February 10 <sup>th</sup>   | Third law of thermodynamics: Nernst heat theorem, statement of concept of residual  |
| February                                     | entropy,  |
| <b>3</b>                                     |   |
| 11 <sup>th</sup> February                    | SUNDAY  |
| Week 5                                       | evaluation of absolute entropy from heat capacity data. Gibbs and Helmholtz   |
| 12 <sup>th</sup> February – 17t <sup>h</sup> | functions; Gibbs function (G) and Helmholtz function (A) as thermodynamic   |
| February                                     | quantities  |
|  |   |
| 18th February                                | SUNDAY  |
| Week 6                                       | A & G as criteria for thermodynamic equilibrium and spontaneity, their  |
| 19 <sup>th</sup> February – 24 <sup>th</sup> | advantage over entropy change. Variation of G and A with P, V and T   |
| February                                     |   |
| 25 <sup>th</sup> February                    | SUNDAY  |
| Week 7                                       | . (ASSIGNMENT-1), TEST  |
| 26thFebruary-02nd                            | Submission of assignments and Queries will be taken.  |
| March  |   |
| 03 <sup>th</sup> March                       | SUNDAY  |
| Week 8                                       | Electrochemistry-III  |
| 04 <sup>th</sup> March-09 <sup>th</sup>      | Electrolytic and Galvanic cells – reversible & Irreversible cells, conventional   |
|  | representation of electrochemical cells.  |
|  |   |
| 10 <sup>th</sup> March                       | SUNDAY  |
| Week 9                                       | EMF of cell and its measurement, Wes ton  |
| 11 <sup>th</sup> March–16 <sup>th</sup>      | standard cell, activity and activity coefficients   |
| March  |   |
| 17 <sup>th</sup> March                       | SUNDAY  |
| XX/1- 10                                     | . Calculation of thermodynamic  |
| Week 10                                      |   |
| 18 <sup>th</sup> March— 22 <sup>th</sup>     | quantities of cell reaction (G, H & K). Types of reversible electrodes – metal- metal ion gas   |
|  | quantities of cell reaction (G, H & K). Types of reversible electrodes – metal- metal ion gas electrode, metal –insoluble salt- anion and redox electrodes. Electrode |
| 18 <sup>th</sup> March– 22 <sup>th</sup>     |   |
| 18 <sup>th</sup> March— 22 <sup>th</sup>     | electrode, metal –insoluble salt- anion and redox electrodes. Electrode   |

| 24 March  | SUNDAY Week               | 11 23-31 March Holi Break                                       |
|---|---------------------------|---|
| Week 12   | , Nernst equations, deriv | ation of cell EMF and single electrode potential.               |
| 01st April – 06th April                         | Standard Hydrogen elect   | rode, reference electrodes, standard electrodes potential, sign |
| _   | conventions, electrochem  | nical se ries and its applications                              |
| 07 <sup>th</sup> April                          | SUNDAY                    |   |
| Week 13   | Concentration cells with  | and without transference, liquid junction potential,            |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April  | application of EMF meas   | surement i.e. valency of ions, solubility product activity      |
|   | 20                        |   |
|   |                           |   |
| 14 <sup>th</sup> April                          | SUNDAY                    |   |
| Week 14   | · *                       | c titration (acid- base and redox). Determination of pH using   |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April | Hydrogen electrode        |   |
|   |                           |   |
| 21 <sup>st</sup> April                          | SUNDAY                    |   |
|   |                           | and glass electrode by potentiometric                           |
|   | methods.                  |   |
| Week 15   | . (ASSIGNMEN              | ,   |
| 22 <sup>nd</sup> April – 27 April               | Submission of assignme    | nts and Queries will be taken.                                  |
| 28 <sup>th</sup> April                          | SUNDAY                    |   |
| Week 16   | Brief Section A, B, C, I  | O Queries/ Doubts   |
| 29 <sup>th</sup> April-30 <sup>th</sup> April   | , , ,                     | -   |

07<sup>th</sup> April

SUNDAY

| [Type here]   |  |  |
|---|--|--|
|   | Summary of Lesson Plan of College Faculty  |  |
| Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2023-24 Semester: Even |  |  |
|   | f. – [A] Poonam [B] Reena  |  |
| Class: B.Sc Pass 4th SEM  | I [Non Med and Med.] Subject- Organic Chemistry  |  |
| 1 <sup>st</sup> January – 15 <sup>t</sup> h   | Practical Slot   |  |
| January   |  |  |
|   |  |  |
| Week 1  | Overview of syllabus, General Reactions Introduction   |  |
| 16 January - 20   |  |  |
| January   |  |  |
| 21st January  | SUNDAY   |  |
| Week 2  | Infrared (IR) absorption spectroscopy  |  |
| 22 <sup>rd</sup> January – 27 <sup>th</sup>   | Molecular vibrations, Hooke's law, selection rules, intensity and position of IR bands,              |  |
| January   |  |  |
| 28 <sup>th</sup> January  | SUNDAY   |  |
| Week 3  | measurement of IR spectrum, fingerprint region, characteristic absorptions of various functional     |  |
| 29 <sup>th</sup> January - 03 <sup>th</sup>   | groups and interpretation of IR spectra of simple organic compounds.                                 |  |
| February<br>04 <sup>th</sup> February   | SUNDAY   |  |
| Week 4  | Applications of IR spectroscopy in structure elucidation of simple organic compounds.                |  |
| 05 <sup>th</sup> February 10 <sup>th</sup>  | (TEST)   |  |
| February 10   | (1231)   |  |
| 11 <sup>th</sup> February   | SUNDAY   |  |
| Week 5  | Amines   |  |
| 12 <sup>th</sup> February – 17t <sup>h</sup>  | Structure and nomenclature of amines, physical properties. Separation of a mixture of primary,       |  |
| February  | secondary and tertiary amines.   |  |
|   |  |  |
| 18th February   | SUNDAY   |  |
| Week 6  | Structural features affecting basicity of amines. Preparation of alkyl and aryl amines (reduction of |  |
| 19 <sup>th</sup> February – 24 <sup>th</sup>  | nitro compounds, nitriles, reductive amination of aldehydic and ketonic compounds.                   |  |
| February  |  |  |
| 25 <sup>th</sup> February   | SUNDAY   |  |
| Week 7  | Gabrielphthalimide reaction, Hofmann bromamide reaction. electrophilic aromatic substitution in      |  |
| 26thFebruary-02nd   | aryl amines, reactions of amines with nitrous acid. (ASSIGNMENT-1) (TEST)                            |  |
| March   |  |  |
| 03 <sup>th</sup> March  | SUNDAY   |  |
| Week 8<br>04 <sup>th</sup> March–09 <sup>th</sup>   | Gabrielphthalimide reaction, Hofmann bromamide reaction. electrophilic aromatic                      |  |
| U4 March-09 M   | substitution in aryl amines, reactions of amines with nitrous acid. (ASSIGNMENT-1)                   |  |
|   | (TEST)   |  |
| 10 <sup>th</sup> March  | SUNDAY   |  |
| Week 9  | reduction of diazonium salts to hyrazines, coupling reaction and its synthetic                       |  |
| 11 <sup>th</sup> March–16 <sup>th</sup>   | application.   |  |
| March   | (TEST)   |  |
| 17 <sup>th</sup> March  | SUNDAY   |  |
| Week 10   | Nitro Compounds  |  |
| 18 <sup>th</sup> March— 22 <sup>th</sup>  |  |  |
| March   | Preparation of nitro alkanes and nitro arenes and their chemical reactions.                          |  |
| 24 March  | SUNDAY Week 11 23-31 March Holi Break  |  |
|   |  |  |
| Week 12   | Mechanism of electrophilic substitution reactions in nitro arenes and their reductions in            |  |
| 01 <sup>st</sup> April – 06 <sup>th</sup> April   | acidic, neutral and alkaline medium.(TEST)   |  |
|   |  |  |

| Week 13   | Aldehydes and Ketones  |
|---|--|
| 8 <sup>th</sup> April – 13 <sup>th</sup> April  | Nomenclature and structure of the carbonyl group. Synthesis of aldehydes and ketones with  |
|   | particular reference to the synthesis of aldehydes from acid chlorides.  |
| 14 <sup>th</sup> April                          | SUNDAY   |
| Week 14   | advantage of oxidation of alcohols with chromium trioxide (Sarett reagent) pyridinium  |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April | chlorochromate (PCC) and pyridinium dichromate., Physical properties. Comparison of  |
|   | reactivities of aldehydes and ketones. Mechanism of nucleophilic additions to carbonyl group with particular emphasis on benzoin, aldol, Perkin and Knoevenagel condensations. |
| 21stApril                                       | SUNDAY   |
| Week 15   | Condensation with ammonia and its derivatives. Wittig reaction. Mannich  |
| 22 <sup>nd</sup> April – 27 April               | reaction.Oxidation of aldehydes, Baeyer–Villiger oxidation of ketones, Cannizzaro  |
|   | reaction. MPV, Clemmensen, Wolff-Kishner, LiAlH4and NaBH4reductions (ASSIGNMENT-   |
|   | 2)   |
|   | Submission of assignments and Queries will be taken.   |
| 28 <sup>th</sup> April                          | SUNDAY   |
| Week 16   | Brief Section A, B, C, D Queries/ Doubts   |
| 29th April-30th April                           |  |

Summary of Lesson Plan of College Faculty Name of College: Pt. Neki Ram Sharma Government College, Rohtak

**Academic Session 2023-24** 

**Semester: Even** 

Name of Asstt./Ass. ProF. –Rekha Gautam, Rinku Lathwal Class: B.Sc Pass 6<sup>th</sup> SEM (Non. Med & Med)

Name of Subject: Inorganic Chemistry

| 1 <sup>st</sup> January – 15 <sup>t</sup> h     | Practical Slot  |
|---|---|
| January   |   |
| Week 1  |   |
| 16 January - 20                                 | Overview of syllabus, General Reactions Introduction                                    |
| January   |   |
| 21st January                                    | SUNDAY  |
| Week 2  | Organometallic Chemistry-   |
| 22 <sup>rd</sup> January – 27 <sup>th</sup>     | Definition, nomenclature and classification of organometallic compounds.                |
|   | Definition, nomenciature and classification of organometatic compounds.                 |
| January   | CYTAID A Y7   |
| 28 <sup>th</sup> January                        | SUNDAY  |
| Week 3  | Preparation, properties, and bonding of alkyls of Li, Al, Hg, and Sn a brief account of |
| 29 <sup>th</sup> January - 03 <sup>th</sup>     | metal-ethylenic complexes.  |
| February  |   |
| 04 <sup>th</sup> February                       | SUNDAY  |
| Week 4  | Mononuclear carbonyls and the nature of bonding in                                      |
| 05 <sup>th</sup> February 10 <sup>th</sup>      | metal carbonyls.  |
| February  | (TEST)  |
| 11 <sup>th</sup> February                       | SUNDAY  |
| Week 5  | Bioinorganic Chemistry-   |
| 12 <sup>th</sup> February – 17t <sup>h</sup>    | Essential and trace elements in biological processes                                    |
| February - 17t                                  | Lossential and trace crements in elelegical processes                                   |
|   | SUNDAY  |
| 18th February                                   | Metalloporphyrins with special  |
| Week 6  |   |
| 19 <sup>th</sup> February – 24 <sup>th</sup>    | reference to haemoglobin and myoglobin.   |
| February  |   |
| 25 <sup>th</sup> February                       | SUNDAY  |
| Week 7  | Biological role of alkali and alkaline earth  |
| 26thFebruary-02nd                               | metal ions with special reference to Ca2+   |
| March   |   |
| 03 <sup>th</sup> March                          | SUNDAY  |
| Week 8  | Nitrogen fixation (ASSIGNMENT-1)  |
| 04 <sup>th</sup> March-09 <sup>th</sup>         |   |
| 10 <sup>th</sup> March                          | SUNDAY  |
| Week 9  | Silicones-  |
| 11 <sup>th</sup> March–16 <sup>th</sup>         | Introduction, preparation, properties, structure and uses.                              |
| March   | introduction, properties, structure and ases.   |
| 17 <sup>th</sup> March                          | SUNDAY  |
|   |   |
| Week 10   | Phosphazenes-   |
| 18 <sup>th</sup> March— 22 <sup>th</sup>        | Introduction, preparation, properties, structure and uses. (TEST)                       |
| March   |   |
| 24 March  | SUNDAY Week 11 23-31 March Holi Break   |
| Week 12   | Acids and Bases, HSAB Concept-  |
|   | Arrhenius, Bronsted – Lowry, the Lux – Flood.   |
| 01st April – 06th April                         |   |
| 07 <sup>th</sup> April                          | SUNDAY  |
| Week 13   | Solvent system and Lewis concepts of  |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April  | acids & bases.  |
| 14 <sup>th</sup> April                          | SUNDAY  |
| Week 14   | Relative strength of acids & bases, Concept of Hard and Soft Acids &                    |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April | Bases   |
| 21stApril                                       | SUNDAY  |
|   |   |
|   |   |
| Week 15   | Symbiosis, electronegativity and hardness and softness. (ASSIGNMENT-2)                  |
| 22 <sup>nd</sup> April – 27 April               | Submission of assignments and Queries will be taken.                                    |
|   |   |
| 28 <sup>th</sup> April                          | SUNDAY  |
|   |   |
|   | 1   |

| Week 16                                       | Brief Section A, B, C, D Queries/ Doubts |
|---|--|
| WCCK 10                                       | Differ Section A, B, C, D Queries Doubts |
| 29 <sup>th</sup> April-30 <sup>th</sup> April |  |
| 29 Aprii-30 Aprii                             |  |
|   |  |

Name of College: Pt. Neki Ram Sharma Government College, Rohtak

**Academic Session 2023-24** 

**Semester: Even** 

Name of Asstt./Ass. Prof. – [A] Neeraj [B] Reena Class: B.Sc Pass 6<sup>th</sup> SEM [Non Med and Med.]

# Paper- XXIII (Theory) Organic Chemistry

| 1 <sup>st</sup> January – 15 <sup>t</sup> h  | Practical Slot  |
|--|---|
| January                                      |   |
| _  |   |
| Week 1                                       | Overview of syllabus, General Reactions Introduction  |
| 16 January - 20                              |   |
| January                                      |   |
| 21st January                                 | SUNDAY  |
| Week 2                                       | Heterocyclic Compounds-I  |
| 22 <sup>rd</sup> January – 27 <sup>th</sup>  | Introduction: Molecular orbital picture and aromatic characteristics of pyrrole, furan, thiophene and |
| January                                      | pyridine.   |
| 28th January                                 | SUNDAY  |
| Week 3                                       | Methods of synthesis and chemical reactions with particular emphasis on the mechanism of              |
| 29 <sup>th</sup> January - 03 <sup>th</sup>  | electrophilic substitution.   |
| February                                     |   |
| 04th February                                | SUNDAY  |
| Week 4                                       | Mechanism of nucleophilic substitution reactions in pyridine derivatives. Comparison of basicity of   |
| 05 <sup>th</sup> February 10 <sup>th</sup>   | pyridine, piperidine and pyrrole.   |
| February                                     | (TEST)  |
| 11th February                                | SUNDAY  |
| Week 5                                       | Heterocyclic Compounds-II   |
| 12 <sup>th</sup> February – 17t <sup>h</sup> | Introduction to condensed five and six- membered heterocycles. Prepration and                         |
| February                                     | reactions of indole, quinoline and isoquinoline with special reference to Fisher                      |
|  | indole synthesis.   |
| 18th February                                | SUNDAY  |
| Week 6                                       | Skraup synthesis and Bischler-Napieralski synthesis, Mechanism of electrophilic substitution          |
| 19 <sup>th</sup> February – 24 <sup>th</sup> | reactions of, quinoline and isoquinoline (TEST)   |
| February                                     |   |
| 25 <sup>th</sup> February                    | SUNDAY  |
| Week 7                                       | Organosulphur Compounds-  |
| 26thFebruary-02nd                            | Nomenclature, structural features, Methods of formation and chemical                                  |
| March  | reactions of thiols, thioethers, sulphonic acids, sulphonamides and                                   |
|  | sulphaguanidine. Synthetic detergents alkyl and aryl sulphonates.                                     |
| 200  | (ASSIGNMENT-1)  |
| 03 <sup>th</sup> March                       | SUNDAY  |
| Week 8                                       | Organic Synthesis via Enolates, Acidity of -hydrogens, alkylation of diethyl malonate and ethyl       |
| 04 <sup>th</sup> March–09 <sup>th</sup>      | acetoacetate. Synthesis of ethyl acetoacetate: the Claisen condensation. Keto-enol tautomerism of     |
|  | ethyl acetoacetate.   |
| 10 <sup>th</sup> March                       | SUNDAY  |
| Week 9                                       | Synthetic Polymers, Addition or chain-growth polymerization. Free radical vinyl polymerization,       |
| 11 <sup>th</sup> March–16 <sup>th</sup>      | ionic vinyl polymerization, Ziegler-Natta polymerization and vinyl polymers.                          |
| March  |   |
| 17 <sup>th</sup> March                       | SUNDAY  |
| Week 10                                      | Condensat ion or step growth polymerization. Polyeste rs ,polyamides, phenol                          |
| 18 <sup>th</sup> March– 22 <sup>th</sup>     | formaldehyde resins, urea formaldehyde resins, epoxy re sins and polyurethanes.                       |
| March  | Natural and synthetic rubbers. (TEST)   |
|  |   |

| 24 March  | SUNDAY Week 11 23-31 March Holi Break   |
|---|---|
| Week 12   | Amino Acids, Peptides& Proteins   |
| 01st April – 06th April                         | Classification, of amino acids. Acid-base behavior, isoelectric point and         |
|   | electrophoresis. Preparation of -amino acids.                                     |
| 07 <sup>th</sup> April                          | SUNDAY  |
| Week 13   | Structure and nomenclature of   |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April  | peptides and proteins. Classification of proteins.                                |
| 14 <sup>th</sup> April                          | SUNDAY  |
| Week 14   | Peptide structure determination, end  |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April | group analysis, selective hydrolysis of peptides                                  |
| 21 <sup>st</sup> April                          | SUNDAY  |
|   |   |
| Week 15   | Classical peptide synthesis, solid—   |
| 22 <sup>nd</sup> April – 27 April               | phase peptide synthesis. Structures of peptides and proteins: Primary & Secondary |
|   | structure. (ASSIGNMENT-2)   |
|   | Submission of assignments and Queries will be taken.                              |
| 28 <sup>th</sup> April                          | SUNDAY  |
|   |   |
| Week 16   | Brief Section A, B, C, D Queries/ Doubts  |
| 29th April-30th April                           |   |

| Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2023-24 Semester: Even |   |  |
|---|---|--|
| Name of Asstt./Ass. Prof. – [A] Bhupender Singh [B] Sandeep   |   |  |
| Class: B.Sc III 6 <sup>th</sup> SEM [Non Med and Med.]  |   |  |
| Name of Subject: Physic   | eal Chemistry   |  |
| 1 <sup>st</sup> January – 15 <sup>th</sup>  | Practical Slot  |  |
| January   |   |  |
|   |   |  |
| Week 1  |   |  |
| 16 January - 20   | Electronic Spectrum   |  |
| January   | Concept of potential energy curves for bonding and antibonding molecular orbitals,  |  |
| 21st January  | SUNDAY  |  |
| Week 2<br>22 <sup>rd</sup> January – 27 <sup>th</sup><br>January  | Qualitative description of selection rules and Franck- Condon principle. Qualitative description of sigma and pie and n molecular orbital (MO) their energy level and respective transitions.   |  |
| 28th January  | SUNDAY  |  |
| Week 3<br>29 <sup>th</sup> January - 03 <sup>th</sup><br>February                                       | Photochemistry Interaction of radiation with matter, difference between thermal and photochemical processes. Laws of photochemistry: Grotthus-Drapper law, Stark- Einstein law (law of photochemical equivalence)                           |  |
| 04th February   | SUNDAY  |  |
| Week 4<br>05 <sup>th</sup> February 10 <sup>th</sup><br>February  | Jablonski diagram depiciting various processes occurring in the excited state, qualitative description of fluorescence, phosphorescence, non-radiative processes (internal conversion, intersystem crossing).                               |  |
| 11th February   | SUNDAY  |  |
| Week 5<br>12 <sup>th</sup> February – 17t <sup>h</sup><br>February                                      | Quantum yield, photosensitized reactions-energy transfer processes (simple examples).   |  |
| 18 <sup>th</sup> February   | SUNDAY  |  |
| Week 6<br>19 <sup>th</sup> February – 24 <sup>th</sup><br>February                                      | <b>Solutions:</b> Dilute Solutions and Colligative Properties. Ideal and non-ideal solutions, methods of expressing concentrations of solutions, activity and activity coefficient.   |  |
| 25 <sup>th</sup> February   | SUNDAY  |  |
| Week 7<br>26t <sup>h</sup> February–02 <sup>nd</sup><br>March   | Dilute solution, Colligative properties, Raolut's law, relative lowering of vapour pressure, molelcular weight determination, Osmosis law of osmotic pressure and its measurement, determination of molecular weight from osmotic pressure. |  |
| 03 <sup>th</sup> March  | SUNDAY  |  |
| Week 8<br>04 <sup>th</sup> March–09 <sup>th</sup>   | Revision, Assignment and Test   |  |
| 10 <sup>th</sup> March  | SUNDAY  |  |
|   |   |  |

| Week 9  | Elevation of boiling point and depression of freezing point, Thermodynamic derivation of relation |
|---|---|
| 11 <sup>th</sup> March–16 <sup>th</sup>         | between molecular weight and elevation in boiling point and depression in freezing point.         |
| March   |   |
| 17th March                                      | SUNDAY  |
| Week 10   | Experimental methods for determining various colligative properties. Abnormal molar mass,         |
| 18 <sup>th</sup> March– 22 <sup>th</sup>        | degree of dissociation and association of solutes.  |
| March   |   |
| 24 March  | SUNDAY  |
| Week 11   | 23-31 March Holi Break  |
| Week 12   | Phase Equilibrium   |
| 01 <sup>st</sup> April – 06 <sup>th</sup>       | Statement and meaning of the terms – phase component and degree of freedom, thermodynamic         |
| April   | derivation of Gibbs phase rule.   |
| 07 <sup>th</sup> April                          | SUNDAY  |
| Week 13   | Phase equilibria of one component system –Example – water and Sulphur systems.                    |
| 8 <sup>th</sup> April – 13 <sup>th</sup> April  |   |
| 14 <sup>th</sup> April                          | SUNDAY  |
| Week 14   | Phase equilibria of two component systems solid-liquid equilibria, simple eutectic Example Pb-    |
| 15 <sup>th</sup> April – 20 <sup>th</sup> April | Ag system, desilverisation of lead.   |
|   |   |
| 21 <sup>st</sup> April                          | SUNDAY  |
|   |   |
| Week 15   | Brief Section A, B, C, D Queries/ Doubts  |
| 22 <sup>nd</sup> April – 27 April               |   |
| 28 <sup>th</sup> April                          | SUNDAY  |
| 20 April  |   |
| Wl-16   | Dei-fC-4 A. D. C. D. O  |
| Week 16   | Brief Section A, B, C, D Queries/ Doubts  |
| 29 <sup>th</sup> April-30 <sup>th</sup> April   |   |
|   |   |